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Synthesis of 1,2-Diazol-4-sulfonamides, 1,2,4- and 1,2,5-Oxadiazol-3-sulfonamides, 1,2,3-Triazol-4-sulfonamides, and Pyrimidine-5-sulfonamides starting from Cyanomethanesulfonyl Chloride

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Summary. Cyanomethanesulfonyl chloride was reacted with amines yielding cyanomethanesulfonamides which could be transformed into alkoxymethylidene and aminomethylidene derivatives. The reaction of alkoxymethylidene derivatives with phenylhydrazine resulted in the formation of 5aminopyrazol-4-sulfonamides, whereas from cyanomethanesulfonamides *via* the *N*-hydroxyamidine derivatives and their reaction with esters 1,2,4-oxadiazol-3-methanesulfonamides became accessible. Nitrosation of cyanomethanesulfonamides yielded 2-hydroxyimino derivatives which were then transformed into 2-hydroxyimino *N*-hydroxyamidine derivatives, and finally cyclized into 4-amino-1,2,5-oxadiazol-3-sulfonamides. On the other hand diazotation of cyanomethanesulfonamides gave the 2-arylhydrazono derivatives, which after transformation into *N*-hydroxyamidine derivatives gave by reaction with POCl₃ 5-amino-1,2,3-triazol-4-sulfonamides. Finally, the reaction between cyanomethanesulfonamides and formamidinium acetate opened an easy access to 4-aminopyrimidine-5-sulfonamides, which could be transformed by trialkyl orthoformiates into substituted pyrimidino[4,5-*e*][1,2,4]thiadiazine derivatives. All intermediates as well as transformation products of the heterocyclic systems were isolated and well characterized. Mechanisms were discussed, and the stereochemistry, when necessary and possible, was elucidated.

Keywords. Cyanomethanesulfonyl chloride; 1,2-Diazol-4-sulfonamides; Oxadiazol-3-sulfonamides; 1,2,3-Triazol-4-sulfonamides; Pyrimidine-5-sulfonamides.

Introduction

Cyanomethanesulfonyl chloride [1] is a highly reactive and unstable compound whose potency for synthesis is only poorly explored. Some reactions with aromatic

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amines and applications in drug research and agriculture chemistry are described [2], the cycloaddition with enamines yielded thietane 1.1-dioxides [3], and by reaction with Et_3N the sulfene was obtained and used in [4+2] cycloadditions [4]. Some years ago, we decided to start a systematic study of this three-functional compound, which should allow reactions at the sulfonyl chloride function, at the methylene group, and at the nitrile function. By a combination of these possibilities we hoped to be able to synthesize a large number of derivatives, and finally to develop easy synthetic routes to different types of heterocyclic sulfonamides showing interesting properties. First results are presented. Some other will be published in a second paper [5].

Results and Discussion

Cyanomethanesulfonyl chloride (1) prepared from chloroacetonitrile by reaction with Na₂SO₃ and PCl₅/POCl₃ [6] reacted with primary or secondary amines yielding the sulfonamides **2b–2m** (Scheme 1). All reactions gave satisfactory to excellent yields when they were performed at 0°C by adding the freshly distilled amine to a solution of **1** in dry(!) Et_2 O. Reasonable yields of the N-unsubstituted **2a** [7] were obtained when a solution of **1** in AcOEt was dropwise neutralized with a saturated solution of NH₃ in CHCl₃. This method seems to us to be more effective than the described one.

The synthesis of sulfonylureas by reaction of **1** with isoureas or cyanamide [8] completely failed, and when **2b** or **2c** in *Et*OH or acetone were refluxed with KNCO [9] no products were isolated. Instead, the sulfonylureas **4a** and **4b** were obtained (yields ~80%) when the starting material **2b** or **2c**, *Bu*Li, and the isocyanate were reacted at -78° C under N₂ in *THF*. We found that this procedure was



Scheme 1

useful not only for the synthesis of the sulfonylureas but even for the acylation of the sulfonamide nitrogen of **2** with acyl chlorides like methyl or benzyl chloroformate yielding the hitherto unknown cyanomethanesulfonyl carbamidates. Some aromatic sulfonyl carbamidates were described and used as agrochemicals [10]. We obtained by this procedure the N-acylated sulfonamides 3a-3p. Compound 3g was obtained from 2c and acetyl chloride, and 3f was prepared from 2c with ethyl chlorothioformate.

The IR spectra (KBr) of 2a-2m are characterized by bands at 3300-3200 (NH), two sharp bands at 2980 and 2920 (CH₂), a low intensity band at 2260-2270 (C \equiv N), and characteristic bands at 1350 and 1150 cm⁻¹ (SO₂). All ¹H NMR spectra show together with the signals caused by the N-substituents a singlet signal around 4.5 ppm caused by the two methylene protons. IR spectra of the sulfony-lureas **4a** and **4b** and of the acylated compounds **3** show an additional carbonyl band at 1740–1720 cm⁻¹. The spectrum of **3f** shows a band at 1675 cm⁻¹ (COS*Et*), and that of **3g** at 1710 cm⁻¹.

The methylene group of the sulfonamides 2 and 3 enabled a number of condensation reactions. Reactions with orthoformates in Ac_2O yielded after refluxing for 2 h the alkoxymethylidene derivatives **5a-50** in good to reasonable yields.



Scheme 2

Compound **5e** was obtained from **5d** by crystallization from isopropanol. The aminomethylidene derivatives **6a–6l** were accessible either by reaction of the parent alkoxymethylidene derivatives **5** with NH₃ in *Et*OH at 0°C (**6e–6g**, **6i**) or by reaction of the methylene sulfonamides **2** with formamidinium acetate in refluxing *Me*OH (Scheme 2).

As can be seen from the spectroscopic data compounds **5** and **6** were isolated as single isomers. The IR spectra of these conjugated nitriles show a band at 2210–2200 cm⁻¹ for the conjugated C \equiv N group, and in their ¹H NMR spectra one sharp singlet between 7.5 and 8.0 ppm indicates the olefinic proton. In the ¹H NMR spectra (400 MHz) of some compounds **6** like **6e** we found a coupling between 2-H and the protons of the amino group. A definitive decision between (*E*)- and (*Z*)-isomer of three-substituted olefins from their spectroscopic data still remains a problem. Estimation of the 2-H signal position using increment systems [11] results only in small differences between (*E*)- and (*Z*)-isomer. The same problem arises using the ³J_{C,H} coupling in the ¹³C NMR spectra [12]. Significant differences are found when calculating the energies of the (*E*)- and (*Z*)-isomers [13]. In all studied examples the (*E*)-isomer was the more stable one. Therefore, we propose that compounds **5** and **6** were obtained as (*E*)-isomers.

The use of these compounds for the synthesis of heterocyclic sulfonamides was first demonstrated by the transformation of **5c** by the reaction with phenylhydrazine in *Et*OH at room temperature yielding the crystalline 1,2-diazol sulfonamide **8b**, yield 70%, from which after hydrolysis by *Et*ONa in *Et*OH at 50°C and acidification **9a** was obtained. The first reaction step may be explained as a *Michael* reaction of the hydrazine to the α,β -unsaturated nitrile followed by an elimination of *Et*OH and an addition of the amino group to the nitrile function [14]. An identical product was isolated from the sequence **5b** \rightarrow **8a** \rightarrow **9a**. The analogous products **9b** and **9c** were prepared from **5g** and **5f** *via* **8c** and **8d** with yields of 38 and 55%. Finally, we transformed **9a** by refluxing with formamidinium acetate in *n-Bu*OH into the bicyclic sulfonamide **10** (Scheme 3).

Transforming the cyanomethanesulfonamides 2 and 3 with $H_2NOH \times HCl$ into the *N*-hydroxyamidines **11a–11h** opened a comfortable route to the 1,2,4-oxadiazol-3-methylsulfonamides. It is essential for the successful reaction, that hydroxylamine is liberated from its salt in an aqueous solution by NaHCO₃, and that a



Scheme 3



Scheme 4

solution of the cyanomethanesulfonamide in *Et*OH is slowly added to that aqueous solution at room temperature [15] (Scheme 4). The *N*-hydroxyamidines **11** were identified by usual spectroscopic and analytical methods. The H–O group gives a sharp band at 3500 cm^{-1} , the H₂N group at 3400 cm^{-1} , and the sulfonamide NH is recorded at 3260 cm^{-1} in the IR spectra. The ¹H NMR spectra of the derivatives with R'=H showed a singlet at ~3.8 ppm from the methylene protons, whereas this signal was found in the spectra of the other compounds at ~4.4 ppm demonstrating the effect of the second substituent.

For characterization and for clarifying the reaction possibilities of 11a-11h we tried a number of reactions with electrophiles, mainly using compound 11b. O-Silylation with ClSi Me_3 in the presence of Et_3N was successful. All expected signals were present in the ¹H NMR spectrum of the crude product, but isolation of the product failed. Reactions with acyl chlorides were successful when done in *THF* or HCCl₃ at room temperature, and in the presence of Et_3N . We obtained the

products 13a, 13d, and 13e from 11b, and 16 from 11h with yields $\sim 80\%$ from reactions with equivalent amounts of benzoyl chloride, methyl, and benzyl chloroformate using this method.

Compound 13b was isolated from the reaction with Ac_2O in AcOH at room temperature, and 13c was prepared by refluxing 11b for 4 h in triethyl orthoformate, yield 60%. When the O-acylated compound 13a was reacted with equivalent amounts of ethyl, phenyl, and 4-nitrophenyl chloroformate the N-acylated derivatives 17a, 17b, and 17c were isolated as crystalline compounds with yields between 50 and 60%. On the other hand, reactions with esters in *Et*OH in the presence of *Et*ONa yielded the 1,2,4-oxadiazol derivatives as is elucidated by the reactions of 11b and 11c with ethyl benzoate yielding 12a and 12b, of 11b with diethyl carbonate to 15, and finally by refluxing of 13b in *Et*OH with *Et*ONa for 2 h yielding 14 (Scheme 4).

The synthesis of 1,2,4-oxadiazol derivatives is possible either by a 1,3-dipolar cycloaddition between a nitrile and a nitriloxide [16] or by the long known cyclization of O-acylated amidoximes, which was first described by *Tiemann* in 1884 [17], but to the best of our knowledge until today none of these methods was used for the synthesis of such sulfonamide derivatives [18] we described here.

Interesting results were found in reactions of the sulfonamides with N-electrophiles (Scheme 5). The reaction of **2b**, **2c**, and **2n** [6] with NaNO₂ and acid at 0°C yielded the (*E*)-oxime derivatives **18a–18c** as crystalline, stable compounds. In analogy to the known synthesis of aminomalononitrile we tried to reduce these compounds to the hitherto unknown α -aminosulfonamides **19** but we failed completely. Whereas α -aminocarboxylic acids and -phosphonic acids and their derivatives are well known compounds the synthesis of α -aminosulfonic acids or their derivatives or other α -N-substituted sulfonyl compounds seems to be difficult or not possible [19].



Scheme 5

The reaction of **3** or **4** with diazonium salts in *Et*OH in the presence of *Ac*ONa enabled the synthesis of the stable hydrazono derivatives 20a-20g as yellow, crystalline compounds with yields from 60 to 80%. The reactions must be done at maximal 0°C, otherwise the mixture becomes red indicating the formation of formazanes. The hydrazone **20h** was prepared from **20f** by refluxing in a solution of *Me*ONa in *Me*OH with 79% yield. It was not accessible by diazotizing of **2c**.

It should be mentioned that a C \equiv N band was not detected in the IR spectra of the hydroxyimino compounds **18a–18c**, whereas in the IR spectra of the hydrazono compounds **20a–20h** the C \equiv N band at 2210 cm⁻¹ was the strongest band of the spectra.

Nevertheless, the hydroxyimino compounds 18a-18c and the hydrazono compounds 20a-20h allowed a number of interesting transformations. It is long known that 1,2-dioximes can be dehydrated forming 1,2,5-oxadiazols (furazans) [20, 21], but there are not many examples described. Therefore, we transformed 18a with H₂NOH×HCl in the presence of NaHCO₃ in H₂O by refluxing for 6 h into the hydroxyamidine derivative 21 which by refluxing with *Et*ONa in *Et*OH or by reaction with POCl₃ gave the 4-amino-1,2,5-oxadiazol sulfonamide 22. When the hydroxyamidine 21 was refluxed with HC(O*Et*)₃, and when the crude intermediate 24 was crystallized from CHCl₃ the *N*-formyl derivative 23 was isolated as a crystalline compound, yield 80%. Refluxing of 21 in Ac_2O ended up with the isolation of the acetylated 25, and finally, the reaction with methyl chloroformate in *THF* gave the N-acylated sulfonamide derivative 26 (Scheme 6).

Formal replacement of the O-atom in 1,2,5-oxadiazols by an N-atom results in 1,2,3-triazols. This structure is well known, and one of the first examples was prepared by *Dimroth* by a reaction between a CH acid with phenyl azide [22]. Another possibility was realized by condensation between a hydrazone and an oxime [23]. As we had successfully used this method for the synthesis of 1,2,5oxadiazols we transferred it to the hydrazono derivatives **20a**, **20b**, **20e**, and **20c**. When they were reacted with H₂NOH×HCl in the presence of NaHCO₃, we obtained the hydroxyimino hydrazono derivatives **27a**–**27d** with yields up to 90% as crystalline compounds. During this reaction the yellow color of the starting materials disappeared. Treatment of **27** with POCl₃ transformed them into the



Scheme 6



1,2,3-triazol-4-sulfonamides **28a–28d** isolated as crystalline compounds with high yields (Scheme 7). These derivatives were treated with *Et*ONa at room temperature, whereby the deacylated derivatives **29a–29c** were obtained. Acylation of the amino group was possible by reaction with Ac_2O as demonstrated by the transformation of **28a** into **31**, or by reaction with ClCOS*Et* as shown by the reaction of **29b** to **32**. Compound **27d** was acetylated to the derivative **30**, but it was not possible to transform this into a triazolo derivative. Furthermore, we have to add that the sulfonylurea derivatives **20f** and **20g** could not be transformed into the hydroxyimino compounds and thereby not transferred into aminotriazolsulfonylurea derivatives. Both decomposed completely during the reaction with H₂NOH×HCl.

The only sparingly soluble 4-aminopyrimidine-5-sulfonamides 33a-33i were obtained from the reaction between the parent aminomethylidene compounds 6 and formamidinium acetate, or with higher yields (~80%) without isolation of the aminomethylidene compounds by refluxing for 6 h of the cyanomethanesulfonamides 2b-2l and an excess of formamidinium acetate in *n*-BuOH (Scheme 8). The *N*-acetyl derivatives 34a-34c were prepared by refluxing of the parent compounds 33 in $AcOH/Ac_2O$. Finally, compounds 33g and 33h were transformed into the pyrimidinothiadiazine derivatives 36a and 36b by refluxing with triethyl orthoformate and catalytic amounts of AcOH [24] for 6 h. Compound 35 was



obtained as a by-product (5%) in the synthesis of **33e** from **2g** and formamidinium acetate.

Experimental Part

Melting points: *Linström* apparatus; IR spectra (KBr): Perkin-Elmer IR 1310, Beckman IR 4240; ¹H and ¹³C NMR spectra: Varian T 60 (¹H, 60 MHz), Bruker WP 80/90 (¹H, 80/90 MHz), WM 300, AM 400 (300/400 MHz for ¹H, 100.614 MHz for ¹³C), room temperature, internal *TMS*, values from 80 MHz spectra in CDCl₃, if not noted otherwise; MS spectra: Finnigan GC MS 4000, MAT 312, MAT 44S; elemental analyses: Institute of Pharmacy or Chemisches Laboratorium, University of Freiburg (Germany) or Greifswald (Germany): All compounds[†] gave satisfactory elemental analyses; results agreed with the calculated values within experimental error. Tetrahydrofuran (*THF*) was stored over CaCl₂ or KOH, then refluxed with Na and benzophenone, and distilled prior to use. Other solvents were dried/purified according to literature procedures [25]. Thin-layer chromatography (tlc): Precoated silica-gel 60F₂₅₄ plates (Merck 5549 or 5715). Column chromatography (CC): Silica gel 60 (Merck 7734).

Cyanomethanesulfonyl chloride (1) See Ref. [1].

Cyanomethanesulfonamide $(2a, C_2H_4N_2O_2S)$ [7]

A satd solution of NH₃ in CHCl₃ was slowly added at 0°C to a solution of 1 (4.2 g, 30 mmol) in 100 cm³ AcOEt. After stirring for 2 h at room temperature, the solvent was evaporated *in vacuo*, the residue was dissolved in AcOEt, and the solution was washed with a satd NH₄Cl solution, dried (Na₂SO₄), and evaporated. Crystallization was induced by a few drops of CHCl₃, and storing at 0°C for several hours. Yield 0.9 g (25%); colorless crystals; mp 97–98°C (AcOEt/CHCl₃); IR: $\bar{\nu} = 3300$,

[†] For simplicity, compounds were named as cyano sulfonamides, ignoring potentially higher-ranking substituents

3220 (NH₂), 2980, 2920 (CH₂), 2260 (CN), 1360, 1140 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): δ = 4.41 (s, CH₂), 6.7–7.1 (s, NH₂) ppm.

General Procedure for the Synthesis of 2

At 0°C, a solution of **1** in 100 cm³ (for 100 mmol) or 20–30 cm³ (for 10 mmol) Et_2 O was slowly added to a solution of the amine (freshly distilled!) in 100–500 (50–60) cm³ Et_2 O. The mixture was stirred for 2 h at room temperature, the precipitate was separated, and the filtrate was concentrated *in vacuo* to the half (only for 100 mmol reactions), washed with H₂O, dried (Na₂SO₄), and evaporated. The residue was crystallized as noted.

N-Phenylcyanomethanesulfonamide (**2b**) See Ref. [6].

N-Benzylcyanomethanesulfonamide (2c, $C_9H_{10}N_2O_2S$)

From **1** (14 g, 100 mmol) and 21.4 g benzylamine (200 mmol). Yield 12.6 g (60%); mp 49–50°C (CHCl₃/CCl₄); IR: $\bar{\nu} = 3290$ (NH), 3090 (CH), 2980, 2930 (CH₂), 2265 (CN), 1340, 1150 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.73$ (s, CH₂), 4.33 (s, CH₂), 5.5–6.0 (s, NH), 7.4 (s, 5 *ar* H) ppm.

N-(4-Methoxybenzyl)cyanomethanesulfonamide (2d, $C_{10}H_{12}N_2O_3S$)

From **1** (1.4 g, 10 mmol) and 2.7 g 4-methoxybenzylamine (20 mmol). Yield 1.2 g (50%); mp 70°C (CHCl₃/CCl₄); IR: $\bar{\nu} = 3300$ (NH), 3010 (CH), 2990, 2930 (CH₂), 2270 (CN), 1335, 1145 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.75$ (s, CH₂), 3.81 (s, OMe), 4.4 (d, J = 6 Hz, CH₂), 5.5 (t, J = 6 Hz, NH), 6.8–7.4 (m, 4 *ar* H) ppm.

Cyanomethanesulfopiperidide (2e, $C_7H_{12}N_2O_2S$)

From **1** (1.4 g, 10 mmol) and 1.7 g piperidine (20 mmol). Yield 1.0 g (60%); mp 134°C (CHCl₃/CCl₄); IR: $\bar{\nu} = 2980$, 2930 (CH₂), 2860 (CH), 2260 (CN), 1340, 1160 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 1.5-1.7$, 3.3–3.5 (2m, 10 *al* H), 4.11 (s, CH₂) ppm.

N-(4-Methoxyphenyl) cyanomethanesulfonamide (**2f**, C₉H₁₀N₂O₃S)

From **1** (1.4 g, 10 mmol) and 2.5 g 4-methoxyaniline (20 mmol). Yield 0.45 g (20%); mp 145°C (CHCl₃/CCl₄); IR: $\bar{\nu} = 3200$ (NH), 2980, 2930 (CH₂), 2830 (CH), 2260 (CN), 1350, 1160 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.85$ (s, *Me*), 4.05 (s, CH₂), 7.0–7.5 (m, 4 *ar* H), 8.6–8.8 (s, NH) ppm.

N-Propylcyanomethanesulfonamide (2g, C₅H₁₀N₂O₂S)

From **1** (1.4 g, 10 mmol) and 1.2 g *n*-propylamine (20 mmol). Purification by Kugelrohr distillation (180°C/0.1 Torr). Yield 0.7 g (45%); mp 40–45°C; IR: $\bar{\nu} = 3280$ (NH), 2980, 2920 (CH₂), 2880 (CH), 2250 (CN), 1325, 1140 (SO₂) cm⁻¹; ¹H NMR: $\delta = 1.0$ (t, J = 8 Hz, Me), 1.4–1.8 (m, CH₂), 3.0–3.4 (q, J = 7 Hz, CH₂), 4.42 (s, CH₂), 6.6–7.0 (t, NH) ppm.

N-(2-*Chlorobenzyl*)*cyanomethanesulfonamide* (**2h**, C₉H₉ClN₂O₂S)

From **1** (7 g, 50 mmol) and 14.2 g 2-chlorobenzylamine (100 mmol). Yield 2.8 g (23%); colorless crystals; mp 67°C; IR: $\bar{\nu}$ = 3314 (NH), 3061 (CH), 2990, 2940 (CH₂), 2258 (CN), 1336, 1141 (SO₂) cm⁻¹; ¹H NMR: δ = 3.8 (s, CH₂), 4.5 (d, *J* = 6.6 Hz, CH₂), 5.7 (t, *J* = 6.6, NH), 7.2–7.5 (m, 4 *ar* H) ppm.

N-(4-Chlorobenzyl)cyanomethanesulfonamide (2i, C₉H₉ClN₂O₂S)

From **1** (7 g, 50 mmol) and 14.2 g 4-chlorobenzylamine (100 mmol). Yield 4.3 g (35%); color-less crystals; mp 90°C; IR: $\bar{\nu} = 3259$ (NH), 2977, 2924 (CH₂), 2264 (CN), 1343, 1144 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.9$ (s, CH₂), 4.4 (d, J = 6.6 Hz, CH₂), 5.7 (t, J = 6.6 Hz, NH), 7.2–7.5 (m, 4 *ar* H) ppm.

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N-(4-Fluorobenzyl)cyanomethanesulfonamide (2k, C₉H₉FN₂O₂S)

From **1** (7 g, 50 mmol) and 17.5 g 4-fluorobenzylamine (100 mmol). Yield 3.0 g (26%); colorless crystals; mp 99°C; IR: $\bar{\nu} = 3290$ (NH), 2988, 2934 (CH₂), 2263 (CN), 1334, 1160 (SO₂) cm⁻¹; ¹H NMR (60 MHz): $\delta = 3.8$ (s, CH₂), 4.3 (d, J = 6.6 Hz, CH₂), 5.7 (t, J = 6.6 Hz, NH), 6.9–7.5 (m, 4 *ar* H) ppm.

N-(4-Bromophenyl)cyanomethanesulfonamide (2l, C₈H₇BrN₂O₂S)

From **1** (7 g, 50 mmol) and 17.2 g 4-bromoaniline (100 mmol). Yield 8.5 g (31%); colorless crystals; mp 107°C (CHCl₃); IR: $\bar{\nu} = 3300$ (NH), 3100 (CH), 2980, 2940 (CH₂), 2280 (CN), 1340, 1150 (SO₂) cm⁻¹; ¹H NMR: $\delta = 4.9$ (s, CH₂), 7.1–7.6 (m, 4 *ar* H), 10.8 (s, NH) ppm.

N-(3,4-Dimethoxyphenyl) cyanomethanesulfonamide (2m, $C_{10}H_{12}N_2O_4S$)

From **1** (3.5 g, 25 mmol) and 7.6 g 3,4-dimethoxyaniline (50 mmol). Yield 1.3 g (10%); mp 120°C (CHCl₃); IR: $\bar{\nu} = 3243$ (NH), 2955, 2914 (CH₂), 2258 (CN), 1349, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 3.73$, 3.75 (2s, 2*Me*), 4.8 (s, CH₂), 6.7–7.0 (m, 3 *ar* H), 10.4 (s, NH) ppm.

Cyanomethanesulfomorpholide (**2n**) See Ref. [6].

General Procedure for the Synthesis of 3

At -78° C, BuLi (6.5 cm³, 10 mmol) was added to a solution of the cyanomethane-sulfonamide **2** (10 mmol) in 50 cm³ *THF*. After 10 min, a solution of the acyl chloride (10 mmol) in 10 cm³ *THF* was added, and the mixture was stirred for 1 h at -78° C. After warming to room temperature, the mixture was hydrolyzed with a satd NaCl solution (100 cm³), the organic layer was separated, dried (MgSO₄), evaporated *in vacuo*, and the residue was crystallized from *Me*OH.

N-Methoxycarbonyl-N-phenylcyanomethanesulfonamide $(3a, C_{10}H_{10}N_2O_4S)$

From **2b** (1.9 g, 10 mmol) and 0.95 g methyl chloroformate. Yield 1.8 g (70%); mp 118°C; IR: $\bar{\nu} = 3070$ (CH), 2980, 2920 (CH₂), 2260 (CN), 1730 (CO), 1370, 1180 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.77$ (s, *Me*), 4.68 (s, CH₂), 7.4 (s, 5 *ar* H) ppm.

N-Benzyl-N-(methoxycarbonyl)cyanomethanesulfonamide (**3b**, C₁₁H₁₂N₂O₄S)

From **2c** (2.1 g, 10 mmol) and 0.95 g methyl chloroformate. Yield 2.2 g (75%); mp 122–123°C; IR: $\bar{\nu} = 3070, 3020$ (CH), 2980, 2920 (CH₂), 2260 (CN), 1730 (CO), 1370, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 3.83$ (s, *Me*), 4.97, 5.00 (2s, 2 CH₂), 7.2–7.5 (s, 5 *ar* H) ppm.

$\textit{N-Benzyl-N-(phenoxycarbonyl)cyanomethanesulfonamide} (3c, C_{16}H_{14}N_2O_4S)$

From **2c** (2.1 g, 10 mmol) and 1.56 g phenyl chloroformate. Yield 2.7 g (75%); mp 105°C; IR: $\bar{\nu} = 3030$ (CH), 2980, 2920 (CH₂), 2270 (CN), 1730 (CO), 1380, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 5.18$ (m, 2 CH₂) 7.1–7.6 (m, 10 *ar* H) ppm.

N-Benzyl-N-(benzyloxycarbonyl)cyanomethanesulfonamide $(3d, C_{17}H_{16}N_2O_4S)$

From **2c** (2.1 g, 10 mmol) and 1.8 g benzyl chloroformate. Yield 2.7 g (80%); mp 91°C; IR: $\bar{\nu} = 3030$ (CH), 2980, 2920 (CH₂), 1740 (CO), 1360, 1160 (SO₂) cm⁻¹; ¹H NMR: $\delta = 4.39$, 4.89, 5.25 (3s, 3 CH₂), 7.5 (s, 10 *ar* H) ppm.

N-(*Benzyloxycarbonyl*)-*N*-(4-methoxybenzyl)cyanomethanesulfonamide (**3e**, $C_{18}H_{18}N_2O_5S$) From **2d** (2.4 g, 10 mmol) and 1.8 g benzyl chloroformate. Yield 3.0 g (80%); mp 103–104°C; IR: $\bar{\nu} = 2980, 2920$ (CH₂), 2250 (CN), 1720 (CO), 1380, 1150 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.76$ (s, *Me*), 4.35, 4.92, 5.26 (3s, 3 CH₂), 6.7–7.4 (m, 9 *ar* H) ppm.

N-Benzyl-N-(S-ethylthiocarbonyl)cyanomethanesulfonamide (3f, $C_{12}H_{14}H_2O_3S_2$)

From **2c** (2.1 g, 10 mmol) and 1.25 g ethyl chlorothioformate. Yield 2.1 g (70%); mp 103–104°C; IR: $\bar{\nu} = 3020$ (CH), 2980, 2920 (CH₂), 2260 (CN), 1675 (COS*Et*), 1360, 1180 (SO₂) cm⁻¹; ¹H NMR: $\delta = 1.3$ (t, J = 7.5 Hz, Me), 2.8–3.1 (q, J = 7.5 Hz, CH₂), 4.52, 5.08 (2s, 2 CH₂), 7.3 (s, 5 *ar* H) ppm.

N-Acetyl-N-benzylcyanomethanesulfonamide (3g, C₁₁H₁₂N₂O₃S)

From **2c** (2.1 g, 10 mmol) and 0.79 g acetyl chloride. Purification by CC (*AcOEt*). Yield 0.5 g (20%); IR (Film): $\bar{\nu} = 2980$, 2920 (CH₂), 2260 (CN), 1710 (CO), 1370, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 2.34$ (s, *Me*), 4.40, 5.11 (2s, 2 CH₂), 7.3 (s, 5 *ar* H) ppm.

N-(*Benzyloxycarbonyl*)-*N*-(4-chlorobenzyl)cyanomethanesulfonamide (**3h**, C₁₇H₁₅ClN₂O₄S) From **2i** (2.4 g, 10 mmol) and 1.8 g benzyl chloroformate. Yield 2.7 g (72%); colorless crystals; mp 90°C; IR: $\bar{\nu} = 3030$ (CH), 2980, 2920 (CH₂), 2260 (CN), 1740 (CO), 1360, 1160 (SO₂) cm⁻¹; ¹H NMR: $\delta = 4.50$, 4.95, 5.3 (3s, 3 CH₂), 7.2–7.5 (m, 9 *ar* H) ppm.

N-(*Benzyloxycarbonyl*)-*N*-(2-chlorobenzyl)cyanomethanesulfonamide (**3i**, C₁₇H₁₅ClN₂O₄S) From **2h** (2.4 g, 10 mmol) and 1.8 g benzyl chloroformate. Yield 2.5 g (66%); mp 78°C; IR: $\bar{\nu} = 3040$ (CH), 2985, 2920 (CH₂), 2260 (CN), 1750 (CO), 1370, 1150 (SO₂) cm⁻¹; ¹H NMR (90 MHz): $\delta = 4.50, 5.10, 5.25$ (3s, 3 CH₂), 7.1–7.45 (m, 9 *ar* H) ppm.

N-(*Benzyloxycarbonyl*)-*N*-(4-fluorobenzyl)cyanomethanesulfonamide (**3k**, $C_{17}H_{15}FN_2O_4S$) From **2k** (2.3 g, 10 mmol) and 1.8 g benzyl chloroformate. Yield 2.2 g (61%); colorless crystals; mp 100°C; IR: $\bar{\nu} = 3060$ (CH), 2985, 2910 (CH₂), 2260 (CN), 1760 (CO), 1370, 1150 (SO₂) cm⁻¹; ¹H NMR (90 MHz): $\delta = 4.40$, 4.90, 5.30 (3s, 3 CH₂), 6.9–7.5 (m, 9 ar H) ppm.

N-(*4*-*Chlorobenzyl*)-*N*-(*methoxycarbonyl*)*cyanomethanesulfonamide* (**3**, C₁₁H₁₁ClN₂O₄S) From **2i** (2.4 g, 10 mmol) and 0.95 g methyl chloroformate. Yield 3.0 g (99%); colorless crystals; mp 145°C; IR: $\bar{\nu} = 2980, 2920$ (CH₂), 2260 (CN), 1740 (CO), 1370, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 3.90$ (s, *Me*), 4.90, 5.10 (2s, 2 CH₂), 7.4 (s, 4 *ar* H) ppm.

N-(2-*Chlorobenzyl*)-*N*-(*methoxycarbonyl*)*cyanomethanesulfonamide* (**3m**, C₁₁H₁₁ClN₂O₄S) From **2h** (2.4 g, 10 mmol) and 0.95 g methyl chloroformate. Yield 0.6 g (20%); colorless crystals; mp 98°C; IR: $\bar{\nu} = 2980$, 2940 (CH₂), 2270 (CN), 1750 (CO), 1370, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 3.90$ (s, *Me*), 5.2 (m, 2 CH₂), 7.35 (s, 4 *ar* H) ppm.

N-(*4-Fluorobenzyl*)-*N*-(*methoxycarbonyl*)*cyanomethanesulfonamide* (**3n**, C₁₁H₁₁FN₂O₄S) From **2k** (2.3 g, 10 mmol) and 0.95 g methyl chloroformate. Yield 1.6 g (60%); colorless crystals; mp 135°C; IR: $\bar{\nu} = 2980, 2920$ (CH₂), 2265 (CN), 1740 (CO), 1370, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 3.90$ (s, *Me*), 5.0 (m, 2 CH₂), 7.0–7.55 (m, 4 *ar* H) ppm.

N-(*4-Bromophenyl*)-*N*-(*methoxycarbonyl*)*cyanomethanesulfonamide* (**30**, C₁₀H₉BrN₂O₄S) From **2l** (2.8 g, 10 mmol) and 0.95 g methyl chloroformate. Yield 1.85 g (67%); colorless crystals; mp 145°C (*Me*OH); IR: $\bar{\nu} = 3000$ (CH₂), 2940 (*Me*), 2270 (CN), 1720 (CO), 1370, 1160 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.75$ (s, *Me*), 4.6 (s, CH₂), 7.2–7.7 (m, 4 *ar* H) ppm.

N-(*Benzyloxycarbonyl*)-*N*-(4-bromophenyl)cyanomethanesulfonamide (**3p**, C₁₆H₁₃BrN₂O₄S) From **2l** (2.8 g, 10 mmol) and 1.8 g benzyl chloroformate. Yield 2.0 g (49%); colorless crystals; mp 159°C (*Me*OH); IR: $\bar{\nu}$ = 3093 (CH), 2925, 2993 (CH₂), 2265 (CN), 1741 (CO), 1373, 1173 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): δ = 5.25, 5.30 (2s, 2 CH₂), 7.5 (m, 9 *ar* H) ppm.

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N-Carbamoyl-N-phenylcyanomethanesulfonamide (4a, C₉H₉N₃O₃S)

From **2b** (1.9 g, 10 mmol) and 1.45 g chlorosulfonyl isocyanate. After stirring for 0.5 h at -78° C the mixture was hydrolyzed with 2.5 cm³ HCl (36%), washed with a satd NaCl solution, the organic layer was separated, dried (MgSO₄), and evaporated *in vacuo*. Yield 1.9 g (80%); mp 140–141°C (CHCl₃/CCl₄); IR: $\bar{\nu}$ = 3440, 3350 (NH₂), 3180 (CH), 2980, 2930 (CH₂), 2270 (CN), 1730 (CO), 1350, 1170 (SO₂) cm⁻¹; ¹H NMR: δ = 5.17 (s, CH₂), 6.1–6.3 (s, NH₂), 7.5 (s, 5 *ar* H) ppm.

N-Benzyl-N-(phenylcarbamoyl)cyanomethanesulfonamide (4b, C₁₆H₁₅N₃O₃S)

From **2c** (2.1 g, 10 mmol) and 1.2 g phenyl isocyanate as described for **4a**. Yield 2.6 g (78%); mp 147°C (CHCl₃/CCl₄); IR: $\bar{\nu}$ = 3360 (NH), 3060 (CH), 2980, 2910 (CH₂), 2270 (CN), 1720 (CO), 1360, 1150 (SO₂) cm⁻¹; ¹H NMR: δ = 3.91, 5.11 (2s, 2 CH₂), 7.0–7.5 (m, 10 *ar* H), 8.4 (s, NH) ppm.

General Procedure for the Synthesis of 5

The trialkyl orthoformate (10 mmol) and the cyanomethanesulfonamide **2** or **3** (10 mmol) in 10 cm^3 Ac_2O were refluxed for 2 h, cooled to room temperature, and evaporated *in vacuo*. The residue was dissolved in CHCl₃, and washed with a NaHCO₃ solution. The organic layer was separated, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified as noted.

(*E*)-*N*-Acetyl-1-cyano-2-methoxy-*N*-phenylethene-1-sulfonamide (**5a**, $C_{12}H_{12}N_2O_3S$) From **2b** (2.0 g, 10 mmol) and 1.1 g HC(OMe)₃. Yield 1.95 g (70%); colorless crystals; mp 122°C (MeOH); IR: $\bar{\nu} = 3050$, 2980 (CH), 2210 (CN), 1690 (CO), 1610 (C=C), 1360, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 1.90$ (s, Me), 4.16 (s, OMe), 7.4 (s, 5 ar H), 8.14 (s, 2-H) ppm.

(*E*)-*N*-Benzyl-1-cyano-2-ethoxy-*N*-(methoxycarbonyl)ethane-1-sulfonamide (**5b**, C₁₄H₁₆N₂O₅S) From **3b** (2.7 g, 10 mmol) and 1.5 g HC(OMe)₃. Yield *ca*. 90%; viscous liquid; IR (Film): $\bar{\nu} = 3060$, 2990, 2980 (CH), 2210 (CN), 1720 (CO), 1610 (C=C), 1370, 1180 (SO₂) cm⁻¹; the product was immediately used for the synthesis of **8a**.

(*E*)-*N*-Benzyl-*N*-(benzyloxycarbonyl)-1-cyano-2-ethoxyethene-1-sulfonamide (**5c**, $C_{20}H_{20}N_2O_5S$) From **3d** (3.4 g, 10 mmol) and 1.5 g HC(OEt)₃. Yield *ca.* 90%; viscous liquid; IR (Film): $\bar{\nu} = 3060$, 2990, 2960 (CH), 2210 (CN), 1720 (CO), 1610 (C=C), 1370, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 1.4$ (t, J = 7 Hz, *Me*), 4.0 (q, J = 7 Hz, CH₂), 5.03, 5.24 (2s, 2 CH₂), 7.3 (m, 10 *ar* H), 7.78 (s, 2-H) ppm; the product was immediately used for the synthesis of **8b**.

(*E*)-*N*-Benzyl-*N*-(benzyloxycarbonyl)-1-cyano-2-methoxyethene-1-sulfonamide (**5d**, C₁₉H₁₈N₂O₅S) From **3d** (2.0 g, 6 mmol) and 1.0 g HC(OMe)₃. Yield 0.4 g (17%); colorless crystals; mp 84–85°C (MeOH); IR: $\bar{\nu} = 3060$ (CH), 2210 (CN), 1720 (CO), 1610 (C=C), 1370, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.80$ (s, Me), 5.00 (s, NCH₂), 5.20 (s, OCH₂), 7.2–7.5 (m, 10 ar H), 7.60 (s, 2-H) ppm.

(*E*)-*N*-*Benzyl*-*N*-(*benzyloxycarbonyl*)-1-*cyano*-2-*isopropoxyethene*-1-*sulfonamide* (**5e**, C₂₁H₂₂N₂O₅S)

From **5d** (2.4 g, 6 mmol) by crystallization from *i-Pr*OH. 0.5 g (20%); colorless crystals; mp 75°C; IR: $\bar{\nu} = 3060, 2990$ (CH), 2210 (CN), 1720 (CO), 1610 (C=C), 1370, 1170 (SO₂) cm⁻¹; ¹H NMR (60 MHz): $\delta = 1.25$ (d, J = 4.4 Hz, Me), 1.40 (d, J = 4.4 Hz, Me), 4.2 (q, CH), 5.0 (s, NCH₂), 5.20 (s, OCH₂), 7.40 (s, 10 *ar* H), 7.90 (s, 2-H) ppm.

(*E*)-*N*-(*Benzyloxycarbonyl*)-*N*-(4-chlorobenzyl)-1-cyano-2-ethoxyethene-1-sulfonamide (**5f**, C₂₀H₁₉ClN₂O₅S)

From **3h** (3.8 g, 10 mmol) and 1.5 g HC(*OEt*)₃. Yield *ca.* 90%; viscous liquid; IR (Film): $\bar{\nu} = 3060$, 2990, 2960 (CH), 2210 (CN), 1720 (CO), 1610 (C=C), 1370, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 1.3$

(t, J = 7 Hz, Me), 4.1 (q, J = 15.4 Hz, CH₂), 4.96 (s, NCH₂), 5.20 (s, OCH₂), 7.2–7.5 (m, 9 *ar* H), 7.8 (s, 2-H) ppm; the product was immediately used for the synthesis of **8d**.

(*E*)-*N*-(*Benzyloxycarbonyl*)-1-cyano-*N*-(4-fluorobenzyl)-2-ethoxyethene-1-sulfonamide (**5g**, $C_{20}H_{19}FN_2O_5S$)

From **3k** (3.6 g, 10 mmol) and 1.5 g HC(O*Et*)₃. Yield *ca*. 90%; ¹H NMR: $\delta = 1.3$ (t, J = 7 Hz, Me), 4.0 (q, J = 15.4 Hz, CH₂), 4.95 (s, NCH₂), 5.20 (s, OCH₂), 6.8–7.5 (m, 9 *ar* H), 7.8 (s, 2-H) ppm; the product was immediately used for the synthesis of **8c**.

(*E*)-*N*-(*Benzyloxycarbonyl*)-*N*-(2-chlorobenzyl)-1-cyano-2-ethoxyethene-1-sulfonamide (**5h**, C₂₀H₁₉ClN₂O₅S)

From **3i** (3.8 g, 10 mmol) and 1.5 g HC(O*Et*)₃. Yield 3.3 g (74%); colorless crystals; mp 100°C (*Et*OH); IR: $\bar{\nu} = 3060, 2990, 2960$ (CH), 2210 (CN), 1720 (CO), 1610 (C=C), 1370, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 1.3$ (t, J = 7 Hz, Me), 4.1 (q, J = 15.4 Hz, CH₂), 5.15, 5.25 (2s, NCH₂, OCH₂), 7.1–7.5 (m, 9 *ar* H), 7.9 (s, 2-H) ppm; ¹³C-NMR: $\delta = 15.11$ (*Me*), 49.12 (NCH₂), 69.62 (OCH₂), 75.09 (OCH₂), 94.6 (C-1), 109.74 (CN), 151.79 (CO), 174.99 (C-2) ppm.

N-(4-Chlorobenzyl)-1-cyano-2-methoxy-N-(methoxycarbonyl)ethene-1-sulfonamide (5i, $C_{13}H_{13}ClN_2O_5S$)

From **31** (3.0 g, 10 mmol) and 1.1 g HC(OMe)₃. Yield *ca*. 90%; yellow viscous liquid; IR (Film): $\bar{\nu} = 3040, 2960$ (CH), 2220 (CN), 1750 (CO), 1610 (C=C), 1360, 1170 (SO₂) cm⁻¹.

$\label{eq:linear} \begin{array}{l} \textit{N-(2-Chlorobenzyl)-1-cyano-2-methoxy-N-(methoxycarbonyl)ethene-1-sulfonamide} \\ \textbf{(5k, $C_{13}H_{13}ClN_2O_5S$)} \end{array}$

From **3m** (3.0 g, 10 mmol) and 1.1 g HC(OMe)₃. Yield *ca*. 50%; yellow viscous liquid; IR (Film): $\bar{\nu} = 3060, 2960$ (CH), 2220 (CN), 1740 (CO), 1620 (C=C), 1380, 1180 (SO₂) cm⁻¹.

$(E)\mbox{-}1\mbox{-}Cyano\mbox{-}N\mbox{-}(4\mbox{-}fluorobenzyl)\mbox{-}2\mbox{-}methoxy\mbox{-}N\mbox{-}(methoxy\mbox{carbonyl})\mbox{ethene-}1\mbox{-}sulfonamide ({\bf 5l}, C_{13}H_{13}FN_2O_5S)$

From **3n** (2.8 g, 10 mmol) and 1.1 g HC(OMe)₃. Yield *ca.* 80%; yellow viscous liquid; IR (Film): $\bar{\nu} = 3040, 2960$ (CH), 2210 (CN), 1750 (CO), 1610 (C=C), 1360, 1170 (SO₂) cm⁻¹.

(*E*)-*N*-(*Benzyloxycarbonyl*)-*N*-(4-bromophenyl)-1-cyano-2-ethoxyethene-1-sulfonamide (**5m**, $C_{19}H_{17}BrN_2O_5S$)

From **3p** (2.1 g, 5 mmol) and 0.75 g HC(OEt)₃. Yield 1.2 g (52%); mp 147°C (*Et*OH); IR: $\bar{\nu} = 3089$, 3061, 2978 (CH), 2234 (CN), 1737 (CO), 1625 (C=C), 1377, 1177 (SO₂) cm⁻¹; ¹H NMR: $\delta = 1.4$ (t, J = 7.7 Hz, Me), 4.25 (q, J = 15.4 Hz, CH₂), 5.2 (s, CH₂), 7.25 (m, 9 *ar* H), 8.0 (s, 2-H) ppm.

(*E*)-*N*-(*Benzyloxycarbonyl*)-*N*-(2-chlorobenzyl)-1-cyano-2-methoxyethene-1-sulfonamide (**5n**, C₁₉H₁₇ClN₂O₅S)

From 4.3 g **5f** by crystallization from *Me*OH. Yield 2.4 g (57%); mp 145°C (*Me*OH); IR: $\bar{\nu} = 3047$, 2958 (CH), 2228 (CN), 1745 (CO), 1617 (C=C), 1363, 1177 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.9$ (s, OMe), 5.15, 5.25 (2s, NCH₂, OCH₂), 7.1–7.5 (m, 9 *ar* H), 7.75 (s, 2-H) ppm.

 $(E)-N-(2-Chlorobenzyl)-1-cyano-2-ethoxy-N-(methoxycarbonyl)ethene-1-sulfonamide (50, C_{14}H_{15}ClN_2O_5S)$

From 3.4 g **5k** by crystallization from *Et*OH. Yield 2.0 g (56%); mp 105°C; IR: $\bar{\nu} = 2980$, 2950 (CH), 2220 (CN), 1740 (CO), 1600 (C=C), 1360, 1170 (SO₂) cm⁻¹; ¹H NMR (90 MHz): $\delta = 1.5$ (t, J = 7.55 Hz, *Me*), 3.8 (s, OM*e*), 4.4 (q, J = 7.55 Hz, OCH₂), 5.1 (s, NCH₂), 7.2–7.5 (m, 4 *ar* H), 8.0 (s, 2-H) ppm.

General Procedures for the Synthesis of 6

Method a. The cyanomethanesulfonamide **2** (10 mmol) and formamidinium acetate (1.0 g, 10 mmol) in $50 \text{ cm}^3 BuOH$ were refluxed for 4–6 h. After cooling to room temperature, the solution was evaporated *in vacuo*, and the residue was crystallized from CHCl₃.

Method b. The 2-alkoxy-1-cyanoethene-1-sulfonamide **5** (5 mmol) was stirred in 300 cm^3 abs. *Et*OH saturated with NH₃ at 0°C for 6–10h. Then, the solvent was evaporated *in vacuo*, and the residue was crystallized from *Me*CN.

(*E*)-2-Amino-N-benzyl-1-cyanoethene-1-sulfonamide (**6a**, C₁₀H₁₁N₃O₃S)

From **2c** (2.1 g), method a. Yield 0.95 g (40%); mp 158–159°C; IR: $\bar{\nu} = 3440, 3300 \text{ (NH}_2), 3070, 3030 \text{ (CH)}, 2200 \text{ (CN)}, 1640 \text{ (C=C)}, 1310, 1160 \text{ (SO}_2) \text{ cm}^{-1}$; ¹H NMR (acetone-d₆): $\delta = 4.2$ (d, J = 6 Hz, CH₂), 6.7 (br t, NH), 7.2–8.0 (m, 5 *ar* H, NH₂, 2-H) ppm.

(*E*)-2-Amino-1-cyano-N-(4-methoxybenzyl)ethene-1-sulfonamide (**6b**, C₁₁H₁₃N₃O₃S)

From **2d** (2.4 g) and 6.0 g formamidinium acetate (60 mmol), method a, 30 min, 100°C, purification by CC (*AcOEt*). Yield 0.5 g (20%) (and 1.6 g **33b**); mp 151–153°C (*AcOEt*); IR: $\bar{\nu} = 3425$, 3300, 3260 (NH), 3210, 3005, 2940, 2840 (CH), 2200 (CN), 1655 (C=C), 1310, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 3.84$ (s, OMe), 4.2 (d, J = 6 Hz, CH₂), 6.5–6.6 (t, J = 6 Hz, NH), 6.8–8.0 (m, 4 *ar* H, NH₂, 2-H) ppm.

$(E) \hbox{-} 2 \hbox{-} Amino \hbox{-} N \hbox{-} phenyl \hbox{-} 1 \hbox{-} cyanoe thene \hbox{-} 1 \hbox{-} sulfonamide (\textbf{6c}, C_9H_9N_3O_2S) \\$

From **2b** (2.0 g), method a. Yield 0.9 g (40%); mp 156°C (CHCl₃); IR: $\bar{\nu}$ = 3430, 3280, 3260 (NH₂), 3200, 3060 (CH), 2200 (CN), 1650 (C=C), 1320, 1170 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): δ = 7.0–8.5 (m, 5 *ar* H, NH₂, 2-H), 9.8–10.2 (br s, NH) ppm.

(*E*)-2-*Amino-N*-(4-chlorobenzyl)-1-cyanoethene-1-sulfonamide (**6d**, $C_{10}H_{10}ClN_3O_2S$) From **2i** (2.4 g), method a. Yield 0.9 g (33%); colorless crystals; mp 156°C; IR: $\bar{\nu} = 3439$, 3295 (NH₂), 2210 (CN), 1655 (C=C), 1310, 1165 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.2$ (d, J = 6.6 Hz, CH₂), 6.8 (t, NH), 7.3–7.95 (m, 4 *ar* H, NH₂, 2-H) ppm.

(*E*)-2-*Amino-N-benzyl-N-(benzyloxycarbonyl)-1-cyanoethene-1-sulfonamide* (**6e**, $C_{18}H_{17}N_3O_4S$) From **5c** (2.0 g), method b. Yield 0.67 g (36%); colorless crystals; mp 155°C; IR: $\bar{\nu} = 3200$ (NH₂), 3070, 3030 (CH), 2210 (CN), 1740 (CO), 1620 (C=C), 1340, 1150 (SO₂) cm⁻¹; ¹H NMR (CDCl₃/*DMSO*-d₆, 400 MHz): $\delta = 4.20$ (d, J = 6 Hz, NCH₂), 5.20 (s, OCH₂), 7.1–7.4 (m, 10 *ar* H), 7.7 (t, J = 6 Hz, 2-H), 8.1 (d, J = 7.5 Hz, 1H, NH₂), 10.3 (s, 1H, NH₂) ppm.

(*E*)-2-*Amino-N-(benzyloxycarbonyl)-N-(4-chlorobenzyl)-1-cyanoethene-1-sulfonamide* (**6f**, C₁₈H₁₆ClN₃O₄S)

From **5f** (2.2 g), method b. Yield 0.5 g (25%); mp 156°C; IR: $\bar{\nu} = 3200$ (NH₂), 3100, 3030 (CH), 2210 (CN), 1720 (CO), 1620 (C=C), 1340, 1150 (SO₂) cm⁻¹; ¹H NMR (CDCl₃/DMSO-d₆): $\delta = 4.2$ (d, J = 5.5 Hz, CH₂), 5.2 (s, CH₂), 7.1–7.5 (m, 9 *ar* H), 8.0 (s, 2-H), 9.8 (s, NH₂) ppm.

$(E)-2-Amino-N-(benzyloxycarbonyl)-N-(4-fluorobenzyl)-1-cyanoethene-1-sulfonamide (6g, C_{18}H_{16}FN_3O_4S)$

From **5g** (2.1 g), method b. Yield 0.5 g (24%); mp 150°C; IR: $\bar{\nu} = 3200$ (NH₂), 3100, 3030 (CH), 2210 (CN), 1720 (CO), 1620 (C=C), 1340, 1150 (SO₂) cm⁻¹; ¹H NMR (CDCl₃/DMSO-d₆): $\delta = 4.2$ (d, J = 5.9 Hz, CH₂), 5.3 (s, CH₂), 7.0–7.5 (m, 9 *ar* H), 7.9 (s, 2-H), 8.7 (s, NH₂) ppm.

(*E*)-2-*Amino-N*-(2-*chlorobenzyl*)-1-*cyanoethene*-1-*sulfonamide* (**6h**, $C_{10}H_{10}ClN_3O_2S$) From **2h** (2.4 g), method a. Yield 0.5 g (18%); mp 123°C; IR: $\bar{\nu} = 3440, 3300$ (NH₂), 3070, 3030 (CH), 2200 (CN), 1640 (C=C), 1310, 1160 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.3$ (d, J = 6.6 Hz, CH₂), 6.8 (t, J = 6.6 Hz, NH), 7.2–8.0 (m, 4 *ar* H, NH₂, 2-H) ppm.

(E)-2-Amino-N-(benzyloxycarbonyl)-N-(2-chlorobenzyl)-1-cyanoethene-1-sulfonamide (**6***i*, C₁₈H₁₆ClN₃O₄S)

From **5h** (2.2 g), method b. Yield 0.5 g (24%); mp 152°C; IR: $\bar{\nu} = 3200 \text{ (NH}_2\text{)}$, 3100, 3030 (CH), 2210 (CN), 1720 (CO), 1620 (C=C), 1340, 1150 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 4.2$ (d, J = 4.4 Hz, CH₂), 5.3 (s, CH₂), 7.2–7.5 (m, 9 ar H), 8.0 (s, 2-H), 8.7 (t, J = 4.4 Hz, 1H, NH₂), 11.7 (s, 1H, NH₂) ppm.

(*E*)-2-*Amino-N*-(4-bromophenyl)-1-cyanoethene-1-sulfonamide (**6**k, C₉H₈BrN₃O₂S)

From **2l** (1.4 g, 5 mmol) in 50 cm³ *Me*OH, method a. Yield 0.5 g (33%); mp 185°C (CHCl₃); IR: $\bar{\nu} = 3440$ (NH₂), 3260 (NH), 2200 (CN), 1650 (C=C), 1310, 1170 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆, 300 MHz): $\delta = 7.0-7.5$ (m, 4 *ar* H), 8.16 (d, J = 11.3 Hz, NH₂), 10.1 (s, 2-H) ppm; MS (70 eV): m/z (%) = 303 (31, M⁺), 172 (86, C₆H₅BrN).

(*E*)-2-Amino-1-cyano-N-(4-fluorobenzyl)ethene-1-sulfonamide (**6**l, C₁₀H₁₀FN₃O₂S)

From **2k** (1.1 g, 5 mmol) in 50 cm³ *Me*OH, method a. Yield 0.6 g (47%); mp 140°C; IR: $\bar{\nu}$ = 3440 (NH₂), 3300 (NH), 2200 (CN), 1650 (C=C), 1310, 1170 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆, 300 MHz): δ = 4.0 (s, CH₂), 7.1–7.4 (m, 4 *ar* H), 7.5 (s, 2-H), 8.0 (s, NH₂) ppm; MS (70 eV): *m*/*z* = 255 (6, M⁺), 124 (100, C₇H₇FN).

(E)-N-Acetyl-2-amino-N-benzyl-1-cyanoethene-1-sulfonamide (7, C₁₂H₁₃N₃O₃S)

Compound **6a** (1.2 g, 5 mmol), 20 cm³ AcOH, and 20 cm³ Ac₂O were refluxed for 3 h, cooled to room temperature, and evaporated *in vacuo*. The residue was dissolved in CHCl₃, washed with an AcONa solution, dried (Na₂SO₄), and the solvent was evaporated *in vacuo*. Yield 0.95 g (70%); mp 145°C (CHCl₃/CCl₄); IR: $\bar{\nu}$ = 3380, (NH₂), 2210 (CN), 1680 (CONH), 1335, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): δ = 2.33 (s, *Me*), 5.0 (s, CH₂), 7.2–8.2 (m, 5 *ar* H, NH₂, 2-H) ppm.

5-Amino-N-benzyl-N-(methoxycarbonyl)-1-phenylpyrazol-4-sulfonamide (8a, C₁₈H₁₈N₄O₄S)

Compound **3b** (2.7 g, 10 mmol) reacted with HC(O*Et*)₃ (1.5 g, 10 mmol) to **5b**. The crude product was dissolved in 30 cm³ *Et*OH, and with stirring, phenylhydrazine (1.1 g, 10 mmol) was slowly added. After 12 h, the precipitate was separated, H₂O was added to the filtrate, and the combined precipitates were crystallized. Yield 2.7 g (70%); mp 172°C (*Me*OH); IR: $\bar{\nu} = 3460$, 3360 (NH₂), 3050, 2980, 2940 (CH), 1720 (CO), 1620 (C=N), 1350, 1180 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.78$ (s, OMe), 4.9–5.1 (m, CH₂, NH₂), 7.2–7.5 (m, 10 *ar* H, 3-H) ppm.

5-Amino-N-benzyl-N-(benzyloxycarbonyl)-1-phenylpyrazol-4-sulfonamide (**8b**, C₂₄H₂₂N₄O₄S) From **5c** (4.0 g, 10 mmol) and 1.1 g phenylhydrazine as described for **8a**. Yield 3.3 g (70%); mp 153°C (*Me*OH); IR: $\bar{\nu}$ = 3450, 3320 (NH₂), 3040, 2960 (CH), 1720 (CO), 1620 (C=N), 1340, 1140 (SO₂) cm⁻¹; ¹H NMR: δ = 4.6–4.9 (br s, NH₂), 5.03, 5.16 (2s, 2 CH₂), 7.2–7.5 (m, 15 *ar* H, 3-H) ppm.

5-Amino-N-(benzyloxycarbonyl)-N-(4-fluorobenzyl)-1-phenylpyrazol-4-sulfonamide (8c, $C_{24}H_{21}FN_4O_4S$)

From **5g** (4.2 g, 10 mmol) and 1.1 g phenylhydrazine as described for **8a**. Yield 3.5 g (73%); mp 140°C (*Me*OH); IR: $\bar{\nu} = 3440$, 3320 (NH₂), 3040, 2960 (CH), 1720 (CO), 1620 (C=N), 1340, 1140 (SO₂) cm⁻¹; ¹H NMR: $\delta = 4.8$ (br s, NH₂), 5.00 (s, NCH₂), 5.20 (s, CH₂), 6.8–7.8 (m, 14 *ar* H, 3-H) ppm.

 $\label{eq:2.1} 5-Amino-N-(benzyloxycarbonyl)-N-(4-chlorobenzyl)-1-phenylpyrazol-4-sulfonamide (8d, C_{24}H_{21}ClN_4O_4S)$

From **5f** (4.3 g, 10 mmol) as described for **8a**. Yield 1.0 g (20%); mp 120°C (*Me*OH); IR: $\bar{\nu} = 3450$, 3320 (NH₂), 3040, 2960 (CH), 1720 (CO), 1620 (C=N), 1340, 1140 (SO₂) cm⁻¹; ¹H NMR: $\delta = 4.8$ (s, NH₂), 4.95 (s, NCH₂), 5.20 (s, OCH₂), 7.1–7.7 (m, 14 *ar* H, 3-H) ppm.

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5-Amino-N-benzyl-1-phenylpyrazol-4-sulfonamide (9a, C₁₆H₁₆N₄O₂S)

A solution of 200 mg Na in 10 cm³ *Et*OH was dropwise added to a suspension of **8a** (3.8 g, 10 mmol) in 50 cm³ *Et*OH, the mixture was stirred at 50°C for 1 h, neutralized with HCl (36%), and evaporated *in vacuo*. The residue was dissolved in CHCl₃, washed with H₂O, dried (Na₂SO₄), and the solvent was evaporated *in vacuo*. Yield 2.3 g (70%); mp 147–148°C (*Me*OH); IR: $\bar{\nu}$ = 3470, 3370 (NH₂), 3100, 2870 (CH), 1620 (C=N), 1310, 1160 (SO₂) cm⁻¹; ¹H NMR: δ = 4.2 (s, CH₂), 4.5–5.2 (br s, NH, NH₂), 7.2–7.7 (m, 10 *ar* H, 3-H) ppm; MS (70 eV): *m/z* (%) = 328 (8.0, M⁺), 159 (100, C₉H₉N₃), 106 (41.8, C₇H₈N), 91 (71.9, C₇H₇).

5-Amino-N-(4-fluorobenzyl)-1-phenylpyrazol-4-sulfonamide (9b, C₁₆H₁₅FN₄O₂S)

From **8c** (0.8 g, 1.8 mmol) as described for **9a**. Yield 0.2 g (33%); mp 190°C (*Me*OH); IR: $\bar{\nu} = 3467$, 3376 (NH₂), 3110, 2872 (CH), 1622 (C=N), 1310, 1157 (SO₂) cm⁻¹; ¹H NMR: $\delta = 4.1$ (d, J = 6.6 Hz, CH₂), 5.00 (s, NH₂), 5.6 (t, J = 6.7 Hz, NH), 6.8–7.5 (m, 9 *ar* H), 7.8 (s, 3-H) ppm; MS (70 eV): m/z (%) = 346 (4, M⁺), 91 (100, C₆H₅N), 124 (54, C₇H₇FN).

5-Amino-N-(4-chlorobenzyl)-1-phenylpyrazol-4-sulfonamide (9c, $C_{16}H_{15}ClN_4O_2S$) From 8d (0.5 g, 1 mmol) as described for 9a. Yield 0.2 g (55%); mp 179°C (*Me*OH); IR: $\bar{\nu}$ = 3463, 3370 (NH₂), 3111, 2867 (CH), 1624 (C=N), 1312, 1161 (SO₂) cm⁻¹; ¹H NMR: δ = 4.1 (d, *J* = 6.7 Hz, CH₂), 5.0 (s, NH₂), 5.8 (t, *J* = 6.7 Hz, NH), 7.0–7.6 (m, 9 *ar* H), 7.8 (s, 3-H) ppm.

2-Benzyl-5-phenylpyrazolo[3,4-e][1,2,4]thiadiazine 1,1-dioxide (10, C₁₇H₁₄N₄O₂S)

Compound **9a** (0.65 g, 2 mmol) and formamidinium acetate (0.45 g, 4 mmol) in 30 cm³ *Bu*OH were refluxed for 2 h, and after cooling to room temperature, the mixture was evaporated *in vacuo*. The residue was dissolved in CHCl₃, washed with H₂O, dried (Na₂SO₄), and the solvent was evaporated *in vacuo*, purification by CC (CHCl₃). Yield 70 mg (10%); mp 202–203°C (CHCl₃); IR: $\bar{\nu} = 3115$, 2960, 2920 (CH), 1580 (C=N), 1310, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 5.03$ (s, CH₂), 7.2–7.8 (m, 10 *ar* H, 3-H), 8.10 (s, 7-H) ppm; MS (70 eV): m/z (%) = 338 (8.5, M⁺), 91 (100, C₇H₇).

2-Amino-N-(4-chlorobenzyl)-2-(hydroxyimino)ethane-1-sulfonamide (11a, C9H12CIN3O3S)

A solution of **2i** (2.4 g, 10 mmol) in 60 cm³ *Et*OH was added dropwise with stirring to a solution of H₂NOH×HCl (1.4 g, 20 mmol) and NaHCO₃ (1.7 g, 20 mmol) in 30 cm³ H₂O, stirring was continued for 12 h, then, the precipitate was separated. Yield 2.2 g (80%); colorless crystals; mp 163°C (*Me*OH); IR: $\bar{\nu} = 3500$, 3400 (NH₂, OH), 3250 (NH), 3080, 2940 (CH), 1660 (C=N), 1310, 1130 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 3.8$ (s, CH₂), 4.2 (d, J = 6.6 Hz, CH₂), 5.45 (s, NH₂), 7.4 (s, 4 *ar* H), 7.8 (t, J = 6.6 Hz, NH), 9.5 (s, OH) ppm.

2-Amino-N-benzyl-2-(hydroxyimino)ethane-1-sulfonamide (11b, C₉H₁₃N₃O₃S)

From **2c** (2.1 g, 10 mmol) as described for **11a**. Yield 2.2 g (90%); mp 169–170°C (*Me*OH); IR: $\bar{\nu} = 3500, 3400$ (NH₂, OH), 3280 (NH), 3080, 2940 (CH), 1660 (C=N), 1310, 1140 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 3.82$ (s, CH₂), 4.1 (d, J = 7 Hz, CH₂), 5.4–5.6 (s, NH₂), 7.3 (s, 5 ar H), 7.6–7.8 (t, J = 7 Hz, NH), 9.5 (s, OH) ppm.

2-Amino-2-(hydroxyimino)-N-phenylethane-1-sulfonamide (11c, C₈H₁₁N₃O₃S)

From **2b** (2.0 g, 10 mmol) as described for **11a**. Yield 2.0 g (87%); mp 120°C (CH₂Cl₂/*Et*OH); IR: $\bar{\nu} = 3490, 3380$ (NH₂, OH), 3250 (NH), 2960 (CH), 1660 (C=N), 1320, 1130 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 3.82$ (s, CH₂), 5.4–5.6 (s, NH₂), 7.0–7.4 (m, 5 *ar* H), 9.5–9.7 (br s, NH, OH) ppm.

2-Amino-2-(hydroxyimino)-N-propylethane-1-sulfonamide (**11d**, C₅H₁₃N₃O₃S)

From **2g** (1.6 g, 10 mmol) as described for **11a**. Yield 1.7 g (90%); mp 130°C (CHCl₃); IR: $\bar{\nu} = 3500$, 3380 (NH₂, OH), 3280 (NH), 2970, 2920, 2870 (CH), 1660 (C=N), 1320, 1160 (SO₂) cm⁻¹; ¹H NMR

(acetone-d₆): $\delta = 0.8-1.0$ (t, J = 7 Hz, Me), 1.4–1.6 (m, CH₂), 2.9–3.1 (t, J = 7 Hz, CH₂), 3.76 (s, CH₂), 5.3 (s, NH₂), 5.9–6.3 (br s, NH), 8.5–9.0 (br s, OH) ppm.

2-Amino-N-benzyl-N-(benzyloxycarbonyl)-2-(hydroxyimino)ethane-1-sulfonamide (11e, C₁₇H₁₉N₃O₅S)

From **3d** (0.7 g, 2 mmol) as described for **11a**. Yield 0.6 g (80%); mp 113°C (CHCl₃); IR: $\bar{\nu} = 3500, 3400$ (NH₂, OH), 3080, 3040, 2960 (CH), 1730 (CO), 1670 (C=N), 1360, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.27, 4.83$ (2s, 2 CH₂), 5.3–5.5 (2s, NH₂, CH₂), 7.3–7.4 (m, 10 *ar* H), 8.9 (s, OH) ppm.

$\label{eq:2-Amino-N-benzyl-2-(hydroxyimino)-N-(methoxycarbonyl) ethane-1-sulfonamide} (11f, C_{11}H_{15}N_3O_5S)$

From **3b** (2.6 g, 10 mmol) as described for **11a**. Yield 2.1 g (70%); mp 110°C (CHCl₃); IR: $\bar{\nu} = 3480$, 3380 (NH₂, OH), 3060, 2940 (CH), 1730 (CO), 1670 (C=N), 1340, 1140 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 3.79$ (s, *Me*), 4.28, 4.82 (2s, 2 CH₂), 5.2–5.4 (s, NH₂), 7.2–7.4 (s, 5 *ar* H), 8.0–10.0 (s, OH) ppm.

$\label{eq:2-Amino-N-benzyl-2-(hydroxyimino)-N-(phenoxycarbonyl) ethane-1-sulfonamide} (11g, C_{16}H_{17}N_3O_5S)$

From **3c** (3.3 g, 10 mmol) as described for **11a**. Yield 2.5 g (70%); mp 163–164°C (CHCl₃); IR: $\bar{\nu} = 3520, 3400$ (NH₂, OH), 3060, 2940 (CH), 1720 (CO), 1670 (C=N), 1360, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.46, 4.98$ (2s, 2 CH₂), 5.2–6.0 (s, NH₂), 7.2–7.6 (m, 10 *ar* H), 7.8–8.1 (s, OH) ppm.

$\label{eq:2-Amino-N-benzyl-N-(S-ethylthiocarbonyl)-2-(hydroxyimino)ethane-1-sulfonamide} (11h, C_{12}H_{17}N_3O_4S_2)$

From **3f** (3.0 g, 10 mmol) as described for **11a**. Yield 2.7 g (80%); mp 122°C (*Me*OH); IR: $\bar{\nu} = 3480-3380$ (NH₂, OH), 3020, 2970, 2930 (CH), 1670 (COS*Et*), 1660 (C=N), 1340, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 1.0-1.3$ (t, J = 8 Hz, *Me*), 2.7–3.0 (q, J = 8 Hz, CH₂), 4.33, 5.03 (2s, 2 CH₂), 5.1–5.5 (s, NH₂), 7.3–7.5 (s, 5 *ar* H, OH) ppm.

$\textit{N-Benzyl-5-phenyl-1,2,4-oxadiazol-3-methanesulfonamide} (12a, C_{16}H_{15}N_3O_3S)$

A solution of 50 mg Na in 10 cm³ *Et*OH was dropwise added to a mixture of **11b** (2.4 g, 10 mmol) and ethyl benzoate (1.9 g, 10 mmol) in 20 cm³ *Et*OH. The mixture was refluxed for 4 h, cooled to room temperature, neutralized with HCl (36%), and evaporated *in vacuo*. H₂O was added to the residue, then, it was extracted with *THF*, the organic layers were dried (MgSO₄), and evaporated *in vacuo*. The residue was dissolved in CHCl₃, and cyclohexane was added until milkiness. After cooling to 0°C for some h, the precipitate was separated. Yield 1.4 g (90%); mp 86°C (*Me*OH); IR: $\bar{\nu} = 3340$ (NH), 3060, 3030 (CH), 2990, 2940 (CH₂), 1610 (C=N), 1340, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.2-4.5$ (m, 2 CH₂), 5.3–5.5 (t, J = 6 Hz, NH), 7.2–8.2 (m, 10 *ar* H) ppm.

N,5-Diphenyl-1,2,4-oxadiazol-3-methanesulfonamide (12b, C₁₅H₁₃N₃O₃S)

From **11c** (1.1 g, 5 mmol) and ethyl benzoate (0.95 g, 5 mmol) as described for **12a**. Yield 1.4 g (90%); mp 133°C (*Me*OH); IR: $\bar{\nu} = 3200$ (NH), 3080 (CH), 2990, 2940 (CH₂), 1600 (C=N), 1340, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.61$ (s, CH₂), 7.0–8.3 (m, 10 *ar* H), 9.0 (br s, NH) ppm.

2-Amino-N-benzyl-2-(benzoyloxyimino)ethane-1-sulfonamide (13a, C₁₆H₁₇N₃O₄S)

Benzoyl chloride (1.4 g, 10 mmol) in 10 cm³ *THF* was slowly added with stirring to a solution of **11b** (2.4 g, 10 mmol) and *Et*₃N (1.0 g, 10 mmol) in 50 cm³ *THF*. Stirring was continued for 2 h, the precipitate (*Et*₃NH⁺Cl⁻) was separated, and the organic phase was extracted with a satd NaCl solution, dried (Na₂SO₄), and evaporated *in vacuo*. Yield 2.7 g (80%); mp 158–160°C (*MeOH*); IR: $\bar{\nu}$ = 3450, 3350 (NH₂), 3210 (NH), 3060, 3020, 2980, 2920 (CH), 1730 (CO), 1620 (C=N), 1320, 1130 (SO₂)

cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.06$ (s, CH₂), 4.3–4.5 (d, J = 6 Hz, CH₂), 6.5–7.0 (s, NH₂), 6.7–7.0 (t, J = 6 Hz, NH), 7.2–8.2 (m, 10 *ar* H) ppm.

2-(Acetoxyimino)-2-amino-N-benzylethane-1-sulfonamide (13b, C₁₁H₁₅N₃O₄S)

Compound **11b** (0.5 g, 2 mmol) in 20 cm³ *Ac*OH and 20 cm³ *Ac*₂O was stirred at room temperature for 3 h. Then the mixture was evaporated *in vacuo*, the residue was dissolved in CHCl₃, washed with H₂O, the organic layer was dried (Na₂SO₄), and evaporated. Yield 0.45 g (80%); mp 138–140°C (*Me*OH); IR: $\bar{\nu} = 3480$, 3350 (NH₂), 3240 (NH), 3040, 2995, 2950 (CH), 1750 (CO), 1640 (C=N), 1320, 1140 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 2.07$ (s, *Me*), 3.90, 4.18 (2s, 2 CH₂), 6.5 (s, NH₂), 7.3 (s, 5 *ar* H), 7.2–8.2 (s, NH) ppm.

2-Amino-N-benzyl-2-{[(bisethoxy)methoxy]imino}ethane-1-sulfonamide (13c, C₁₄H₂₃N₃O₅S)

Compound **11b** (0.9 g, 4 mmol) in 20 cm³ HC(O*Et*)₃ was refluxed for 4 h. After cooling to room temperature, the mixture was concentrated *in vacuo*, the residue was dissolved in CHCl₃, and after addition of *n*-pentane the product was separated. Yield 0.8 g (60%); mp 72°C (toluene); IR: $\bar{\nu} = 3480$, 3380 (NH₂), 3260 (NH), 3090, 2990, 2950, 2900 (CH), 1660 (C=N), 1320, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 1.0-1.2$ (t, J = 7 Hz, 2*Me*), 3.5–3.8 (q, J = 7 Hz, 2 CH₂), 3.87, 4.35 (2s, 2 CH₂), 5.4–5.7 (s, NH₂), 6.5–6.8 (s, NH), 7.2–7.5 (m, 5 *ar* H, CH) ppm.

$\label{eq:2-Amino-N-benzyl-2-(methoxycarbonyloxyimino) ethane-1-sulfonamide~(13d,~C_{11}H_{15}N_3O_5S)$

HC(OMe)₃ (0.4 g, 4 mmol) in 10 cm³ CHCl₃ was dropwise added to a mixture of **11b** (0.9 g, 4 mmol) and Et_3N (0.4 g, 4 mmol) in 10 cm³ CHCl₃. The mixture was refluxed for 4 h, cooled to room temperature, the precipitate was separated, and the organic solvent was evaporated *in vacuo*. Yield 0.8 g (80%); mp 122–123°C (toluene); IR: $\bar{\nu} = 3480$, 3360 (NH₂), 3250 (NH), 3015, 2990, 2930 (CH), 1750 (CO), 1310, 1130 (SO₂) cm⁻¹; ¹H NMR: $\delta = 3.86$ (s, OMe), 3.97 (s, CH₂), 4.3–4.4 (d, J = 7 Hz, CH₂), 5.3–5.7 (s, t, NH₂, NH), 7.4–7.5 (s, 5 *ar* H) ppm.

$2\mbox{-}Amino\mbox{-}N\mbox{-}benzyl\mbox{-}2\mbox{-}(benzyl\mbox{-}xycarbonyl\mbox{-}xyimino\mbox{-})ethane\mbox{-}1\mbox{-}sulfonamide\mbox{-}(13e,\mbox{-}C_{17}H_{19}N_3O_5S)$

From **11b** (2.4 g, 10 mmol) and benzyl chloroformate (1.7 g, 10 mmol) as described for **13a**. Yield 3.0 g (80%); mp 155°C (toluene); IR: $\bar{\nu} = 3460$, 3350 (NH₂), 3240 (NH), 1760 (CO), 1630 (C=N), 1310, 1140 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 3.9$ (s, CH₂), 4.3 (d, CH₂), 5.2 (s, CH₂), 6.5–6.7 (s, NH₂), 7.4 (m, 10 *ar* H), 8.6–8.8 (s, NH) ppm.

N-Benzyl-5-methyl-1,2,4-oxadiazol-3-methanesulfonamide (14, C₁₁H₁₃N₃O₃S)

A solution of 100 mg Na in 10 cm³ *Et*OH was dropwise added to a suspension of **13b** (1.4 g, 5 mmol) in 50 cm³ *Et*OH, the mixture was refluxed for 2 h, cooled to room temperature, and evaporated *in vacuo*. The residue was dissolved in a few cm³ H₂O, neutralized with HCl (36%), and extracted with CHCl₃. The combined organic layers were dried (Na₂SO₄), and evaporated *in vacuo*. Yield 1.1 g (80%); mp 74°C (*Me*OH); IR: $\bar{\nu} = 3240$ (NH), 2940, 2980 (CH₂), 1580 (C=N), 1320, 1140 (SO₂) cm⁻¹; ¹H NMR: $\delta = 2.54$ (s, *Me*), 4.33, 4.3–4.5 (s, d, 2 CH₂), 5.2–5.4 (t, J = 7 Hz, NH), 7.4 (s, 5 *ar* H) ppm.

N-Benzyl-4,5-dihydro-5-oxo-1,2,4-oxadiazol-3-methanesulfonamide (**15**, $C_{10}H_{11}N_3O_4S$) From **11b** (0.5 g, 2 mmol) and diethyl carbonate (0.24 g, 2 mmol) as described for **12a**. Yield 0.45 g (80%); mp 149°C (CHCl₃/cyclohexane); IR: $\bar{\nu} = 3400$, 3360, 3300, (NH), 2980, 2920 (CH₂), 1810, 1790 (CO), 1600 (C=N), 1330, 1160 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.37$ (d, CH₂), 4.4 (s, CH₂), 7.0–8.5 (s, 5 *ar* H, 2 NH) ppm.

2-Amino-N-benzyl-2-(benzyloxycarbonyloxyimino)-N-(S-ethylthiocarbonyl)ethane-1-sulfonamide (16, C₂₀H₂₃N₃O₆S₂)

From **11h** (1.65 g, 5 mmol) and benzyl chloroformate (0.85 g, 5 mmol) as described for **13a**. Yield 1.8 g (80%); mp 74°C (*Me*OH); IR: $\bar{\nu} = 3480$ (NH), 3050, 2840 (CH), 1660 (COS*Et*), 1310, 1125 (SO₂)

cm⁻¹; ¹H NMR: $\delta = 1.1-1.3$ (t, J = 7 Hz, Me), 2.8–3.1 (q, J = 8 Hz, CH₂), 4.31, 4.88, 5.15 (3s, 3 CH₂), 5.3–5.5 (s, NH₂), 7.2–7.4 (m, 10 *ar* H) ppm.

$\label{eq:2-Amino-2-(benzoyloxyimino)-N-benzyl-N-(ethoxycarbonyl)ethane-1-sulfonamide} (17a, C_{19}H_{21}N_3O_6S)$

From **13a** (1.1 g, 3 mmol) and ethyl chloroformate (0.35 g, 3 mmol) as described for **3a**. Purification by CC (CH₂Cl₂/*AcOEt*). Yield 0.75 g (60%); mp 157–158°C (CH₂Cl₂/*n*-pentane); IR: $\bar{\nu}$ = 3400 (NH₂), 1720 (CO), 1640 (C=N), 1360, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): δ = 1.1–1.4 (t, *J* = 7 Hz, *Me*), 4.1–4.5 (q, *J* = 8 Hz, CH₂), 4.56, 4.93 (2s, 2 CH₂), 6.0–6.2 (m, NH₂), 7.0–8.3 (m, 10 *ar* H) ppm.

$\label{eq:2-Amino-2-(benzoyloxyimino)-N-benzyl-N-(phenoxycarbonyl)ethane-1-sulfonamide} (17b, C_{23}H_{21}H_3O_6S)$

From **13a** (3.4 g, 10 mmol) and phenyl chloroformate (1.56 g, 10 mmol) as described for **13a**. Yield 2.8 g (60%); mp 156–157°C (*Me*OH); IR: $\bar{\nu}$ = 3460, 3380 (NH₂), 1725 (CO), 1630 (C=N), 1360, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): δ = 4.65, 5.07 (2s, 2 CH₂), 6.3–6.6 (s, NH₂), 7.0–8.3 (m, 15 *ar* H) ppm.

$\label{eq:2-Amino-2-(benzoyloxyimino)-N-benzyl-N-(4-nitrophenoxycarbonyl) ethane-1-sulfonamide} (17c, C_{23}H_{20}N_4O_8S)$

From **13a** (1.7 g, 5 mmol) and 4-nitrophenyl chloroformate (1.0 g, 5 mmol) as described for **13a**. Purification by CC (CHCl₃/*AcOEt* = 10/1). Yield 1.3 g (50%); mp 78–81°C (CHCl₃/*AcOEt*); IR: $\bar{\nu}$ = 3480, 3380 (NH₂), 3080, 3030, 2920 (CH), 1730 (CO), 1640 (C=N), 1520 (NO₂), 1350, 1160 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): δ = 4.45, 5.1 (2s, 2 CH₂), 5.2–5.4 (s, NH₂), 7.4–8.5 (m, 14 *ar* H) ppm.

N-Benzyl-cyano(hydroxyimino)methanesulfonamide (18a, C₉H₉N₃O₃S)

An ice-cold solution of NaNO₂ (2.1 g, 30 mmol) in 50 cm³ H₂O was added with vigorous stirring and cooling to a solution of **2c** (2.1 g, 10 mmol) in 30 cm³ *Ac*OH. After stirring at room temperature for 12 h 1.25 cm³ HCl (36%) were added, the mixture was extracted with *THF*, the organic layer was dried (MgSO₄), and evaporated *in vacuo*. The residue was dissolved in a few cm³ CHCl₃, and stored at 0°C. Yield 1.4 g (58%); mp 108–109°C (CHCl₃); IR: $\bar{\nu} = 3400$, 3280 (OH, NH), 3070 (CH), 1330, 1165 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.4$ (d, J = 6 Hz, CH₂), 7.3 (s, 5 *ar* H), 7.9–8.1 (t, NH), 15.9 (s, OH) ppm.

Cyano(hydroxyimino)-N-phenylmethanesulfonamide (18b, C₈H₇N₃O₃S)

From **2b** (2.0 g, 10 mmol) as described for **18a**. Yield 1.4 g (65%); mp 128°C (*Me*OH); IR: $\bar{\nu} = 3260$ (OH, NH), 1370, 1180 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 7.3$ (s, 5 *ar* H), 9.6–9.8 (s, NH), 13.5–14.0 (s, OH) ppm.

Cyano(hydroxyimino)methanesulfonylmorpholide (**18c**, C₆H₉N₃O₄S)

From **2n** (1.9 g, 10 mmol) as described for **18a**. Yield 1.3 g (60%); mp 142–146°C (dec, *Me*OH); IR: $\bar{\nu} = 3420$ (OH), 2920, 2800 (CH), 1360, 1180 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 3.2-3.5$ (m, 2 CH₂), 3.6–3.8 (m, 2 CH₂), 14.0–14.02 (s, OH) ppm.

$\label{eq:2.1} N-Benzyl-N-(benzyloxycarbonyl) cyano(4-methylphenylhydrazono) methane-sulfonamide~(20a,~C_{24}H_{22}N_4O_4S)$

At 0°C, a solution of NaNO₂ (0.7 g) in 3 cm³ H₂O was added with stirring to a solution of *p*-toluidine (1.1 g, 10 mmol) in 3.5 cm³ HCl (36%). This solution was dropwise added with stirring to a suspension of **3d** (3.45 g, 10 mmol) and *Ac*ONa (4.55 g) in 8 cm³ H₂O at max 5°C. After stirring for 12 h at room temperature, the precipitate was separated and dried *in vacuo*. Yield 3.2 g (70%); mp 118°C (*Me*OH); IR: $\bar{\nu} = 3230$ (NH), 3030 (CH), 2200 (CN), 1740 (CO), 1530 (C=N-NH–), 1380, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 2.31$ (s, *Me*), 5.00 (s, CH₂), 5.2 (s, CH₂), 7.0–7.4 (m, 14 *ar* H), 9.8 (s, NH) ppm.

 $\label{eq:linear} N-(Benzyloxycarbonyl) cyano-N-(4-methoxybenzyl)-(4-methyl-phenylhydrazono) methanesulfonamide (20b, C_{25}H_{24}N_4O_5S)$

From **3e** (3.75 g, 10 mmol) and *p*-toluidine (1.1 g, 10 mmol) as described for **20a**. Yield 3.9 g (80%); mp 134–135°C (*Me*OH); IR: $\bar{\nu} = 3220$ (NH), 3050, 2930 (CH), 2200 (CN), 1740 (CO), 1530 (C=N-NH-), 1380, 1165 (SO₂) cm⁻¹; ¹H NMR: $\delta = 2.33$ (s, *Me*), 3.73 (s, *OMe*), 4.97 (s, CH₂), 5.26 (s, CH₂), 6.8–7.5 (m, 13 *ar* H), 9.4 (s, NH) ppm.

N-Benzyl-N-(benzyloxycarbonyl)cyano(phenylhydrazono)methanesulfonamide

 $(20c, C_{23}H_{20}N_4O_4S)$

From **3d** (3.45 g, 10 mmol) and aniline (1.0 g, 10 mmol) as described for **20a**. Yield 3.1 g (70%); mp 137°C (*Me*OH); IR: $\bar{\nu} = 3240$ (NH), 3060 (CH), 2200 (CN), 1760 (CO), 1530 (C=N–NH–), 1385, 1165 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 5.07$ (s, CH₂), 5.28 (s, CH₂), 7.1–7.5 (m, 15 *ar* H), 11.5 (s, NH) ppm.

$\label{eq:cyano-N-(methoxycarbonyl)(4-methylphenylhydrazono)-N-phenylmethanesulfonamide} (20d, C_{17}H_{16}N_4O_4S)$

From **3a** (2.5 g, 10 mmol) and *p*-toluidine (1.1 g, 10 mmol) as described for **20a**. Yield 2.2 g (60%); mp 151°C (*Me*OH); IR: $\bar{\nu}$ = 3180 (NH), 2950 (CH), 2200 (CN), 1720 (CO), 1540 (C=N–NH–), 1380, 1180 (SO₂) cm⁻¹; ¹H NMR: δ = 2.38 (s, *Me*), 3.85 (s, *OMe*), 7.2–7.8 (m, 9 *ar* H), 12.8 (s, NH) ppm.

$\label{eq:n-Benzylcyano-N-(methoxycarbonyl)-4-(methylphenylhydrazono)} methanesulfonamide \mbox{(20e, $C_{18}H_{18}N_4O_4S$)}$

From **3b** (2.6 g, 10 mmol) and *p*-toluidine (1.1 g, 10 mmol) as described for **20a**. Yield 2.8 g (70%); mp 140°C (*Me*OH); IR: $\bar{\nu} = 3200$ (NH), 3030, 2950 (CH), 2200 (CN), 1730 (CO), 1540 (C=N–NH–), 1370, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 2.30$ (s, *Me*), 3.81 (s, OM*e*), 5.00 (s, CH₂), 7.0–7.5 (m, 9 *ar* H), 9.5–10.0 (s, NH) ppm.

N-Benzylcyano-N-(phenylcarbamoyl)(phenylhydrazono)methanesulfonamide (**20f**, $C_{22}H_{19}N_5O_3S$) From **4b** (3.3 g, 10 mmol) and aniline (1.0 g, 10 mmol) as described for **20a**. Yield 3.3 g (78%); mp 132°C (*MeOH*); IR: $\bar{\nu} = 3370$ (NH), 3070, 2970 (CH), 2200 (CN), 1710 (CO), 1530 (C=N-NH-), 1370, 1160 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 5.07$ (s, CH₂), 7.0–7.6 (m, 15 *ar* H), 9.2 (s, NH), 11.4 (s, NH) ppm.

N-*Carbamoylcyano-N-phenyl(phenylhydrazono)methanesulfonamide* (**20**g, C₁₅H₁₃N₅O₃S) From **4a** (2.4 g, 10 mmol) and aniline (1.0 g, 10 mmol) as described for **20a**. Yield 2.9 g (72%); mp 131°C (*Me*OH); IR: $\bar{\nu}$ = 3470, 3360 (NH), 3010, 2970 (CH), 2205 (CN), 1700 (CO), 1530 (C=N-NH-), 1370, 1180 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): δ = 6.8 (s, NH₂), 7.2–7.7 (m, 10 *ar* H) 11.4 (s, NH) ppm.

N-Benzylcyano-(phenylhydrazono)methanesulfonamide (20h, $C_{15}H_{14}N_4O_2S$)

A solution of 0.45 g Na in 50 cm³ *Me*OH was slowly added to a solution of **20f** (4.3 g, 10 mmol) in 50 cm³ *Me*OH. The mixture was refluxed for 6 h, hydrolyzed with HCl (36%), and evaporated *in vacuo*. The residue was dissolved in *Et*₂O, washed with H₂O, dried (Na₂SO₄), and the solvent was evaporated *in vacuo*. Yield 2.5 g (79%); mp 108°C (CHCl₃/CCl₄); IR: $\bar{\nu}$ = 3250 (NH), 3070 (CH), 2200 (CN), 1330, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): δ = 4.4 (s, CH₂), 7.1–7.5 (m, 10 *ar* H), 7.3–7.7 (s, NH), 10.4–11.1 (s, NH) ppm.

2-Amino-N-benzyl-1,2-bis(hydroxyimino)ethane-1-sulfonamide (21, C₉H₁₂N₄O₄S)

From **18a** (2.4 g, 10 mmol) and H₂NOH×HCl (1.4 g, 20 mmol) as described for **2i**. The viscous residue was purified by CC (*AcOEt*). Yield 1.3 g (50%); mp 120°C (*AcOEt*); IR: $\bar{\nu}$ = 3480, 3390 (NH₂, OH),

3270 (NH), 3010, 2930 (CH), 1350, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): δ = 4.33 (d, *J* = 7 Hz, CH₂), 5.7–5.9 (br s, NH₂), 6.9–7.1 (t, NH), 7.2–7.5 (m, 5 *ar* H), 9.0–16.0 (br s, 2 OH) ppm.

4-Amino-N-benzyl-1,2,5-oxadiazol-3-sulfonamide (22, C₉H₁₀N₄O₃S)

Method a. A solution of 0.2 g Na in 10 cm³ *Et*OH was slowly added to a suspension of **21** (1.35 g, 5 mmol) in 20 cm³ *Et*OH, the mixture was refluxed for 4 h, cooled to room temperature, neutralized with HCl (36%), and evaporated *in vacuo*. The residue was dissolved in CHCl₃, washed with H₂O, dried (Na₂SO₄), and evaporated *in vacuo*. The residue was purified by CC (CHCl₃/*AcOEt* = 5/1).

Method b. Compound **21** (0.27 g, 1 mmol) in $20 \text{ cm}^3 \text{ POCl}_3$ was stirred at room temperature for 2 h, the solvent was evaporated *in vacuo*, the residue was dissolved in *THF*, washed with a satd NaCl solution, dried (Na₂SO₄), and the solvent was evaporated *in vacuo*.

Yield a. 0.6 g (50%), b. 0.1 g (40%); mp 76°C (CHCl₃/AcOEt); IR: $\bar{\nu}$ = 3480, 3340, 3300 (NH₂, NH), 3060, 2920 (CH), 1630 (C=N), 1350, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): δ = 4.5 (d, J = 6 Hz, CH₂), 5.3–5.9 (br s, NH₂), 7.4 (s, 5 *ar* H), 7.9–8.1 (br s, NH) ppm; MS (70 eV): m/z (%) = 255 (9.56, M⁺+1), 91 (100, C₇H₇⁺).

N-Benzyl-4-formylamino-1,2,5-oxadiazol-3-sulfonamide (23, C₁₀H₁₀N₄O₄S)

The crude **24** was dissolved in CHCl₃, the solution was washed with H₂O, dried (Na₂SO₄), and the solvent was evaporated *in vacuo*. Yield 1.15 g (80%); mp 84–85°C (CHCl₃); IR: $\bar{\nu} = 3340$ (NH), 3090, 2890, 2870 (CH), 1700 (CO), 1360, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.3$ (s, CH₂), 7.2 (s, 5 *ar* H), 7.5–8.0 (br s, NH), 8.6–8.8 (br s, NH, CHO) ppm; MS (70 eV): m/z (%) = 283 (13.91, M⁺), 91 (100, C₇H₇⁺).

N-Benzyl-4-(ethoxymethylimino)-1,2,5-oxadiazol-3-sulfonamide (24, C₁₂H₁₄N₄O₄S)

Compound **21** (1.35 g, 5 mmol) in 20 cm³ HC(O*Et*)₃ was refluxed for 4 h, formed *Et*OH was distilled off, then the mixture was cooled to room temperature, and the solvent was evaporated *in vacuo*. The crude product, ¹H NMR: $\delta = 1.2-1.5$ (t, J = 7 Hz, Me), 4.0–4.5 (m, 2 CH₂), 5.9–6.1 (br s, NH), 7.2–7.4 (s, 5 *ar* H), 8.12 (s, N=CH) ppm, was immediately used for the synthesis of **23**.

4-Acetylamino-N-benzyl-1,2,5-oxadiazol-3-sulfonamide (25, C₁₁H₁₂N₄O₄S)

Compound **21** (1.35 g, 5 mmol) in 30 cm³ Ac_2O was refluxed for 4 h, cooled to room temperature, concentrated *in vacuo*, dissolved in CHCl₃, washed with H₂O, dried (Na₂SO₄), and the solvent was evaporated *in vacuo*. Yield 1.35 g (80%); mp 90°C (CHCl₃); IR: $\bar{\nu} = 3340$ (NH), 3040, 2980, 2930 (CH), 1720 (CO), 1360, 1150 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 2.28$ (s, *Me*), 4.5 (s, CH₂), 7.3 (s, 5 *ar* H), 7.6–7.9 (br s, NH). 9.3–9.6 (br s, NH) ppm.

4-Amino-N-benzyl-N-(methoxycarbonyl)-1,2,5-oxadiazol-3-sulfonamide (26, C₁₁H₁₂N₄O₅S)

Under N₂ and with stirring, methyl chloroformate (0.2 g, 2 mmol) in 5 cm³ *THF* was added to a solution of **21** (0.54 g, 2 mmol) in 50 cm³ *THF*, stirring was continued for 30 min, then Et_3N (0.4 g, 4 mmol) and additional methyl chloroformate (0.2 g, 2 mmol) in 5 cm³ *THF* were added. The mixture was refluxed for 1 h, cooled to room temperature, washed with a satd NaCl solution, dried (MgSO₄), and evaporated *in vacuo*. The residue was purified by CC (CHCl₃). Yield 0.3 g (50%); mp 110–111°C (CHCl₃); IR: $\bar{\nu}$ = 3490, 3360 (NH₂), 1720 (CO), 1330, 1140 (SO₂) cm⁻¹; ¹H NMR: δ = 3.76 (s, OMe), 4.6–5.1 (br s, NH₂, CH₂), 7.4 (m, 5 *ar* H) ppm; MS (70 eV): *m/z* (%) = 313 (5.16, M⁺+1), 164 (100, C₉H₁₀NO₂).

$\label{eq:2-Amino-N-benzyl-N-(benzyloxycarbonyl)-2-hydroxyimino-1-(4-methylphenylhydrazono) ethane-1-sulfonamide~(27a, C_{24}H_{25}N_5O_5S)$

From **20a** (4.6 g, 10 mmol) and 1.4 g H₂NOH×HCl as described for **11a**. Yield 4.4 g (90%); mp 109°C (*Me*OH); IR: $\bar{\nu} = 3460-3260$ (NH₂, NH, OH), 3030 (CH), 1700 (CO), 1360, 1160 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 2.28$ (s, *Me*), 5.03 (s, CH₂), 5.13 (s, CH₂), 5.8–6.0 (br s, NH₂), 7.0–7.6 (m, 14 *ar* H, OH), 12.9–13.1 (br s, NH) ppm.

2-Amino-N-(benzyloxycarbonyl)-2-hydroxyimino-N-(4-methoxybenzyl)-1-

 $(4\mbox{-}methylphenylhydrazono) ethane-1\mbox{-}sulfonamide~({\bf 27b},\,C_{25}H_{27}N_5O_6S)$

From **20b** (4.9 g, 10 mmol) and 1.4 g H₂NOH×HCl as described for **11a**, 2 h at 50°C, 12 h of room temperature. Yield 4.4 g (85%); mp 114–116°C (*Me*OH/H₂O = 9/1); IR: $\bar{\nu}$ = 3520, 3420, 3300 (NH), 3050, 2950, 2850 (CH), 1720 (CO), 1620 (C=N), 1520 (C=N–NH–) 1340, 1170 (SO₂) cm⁻¹; ¹H NMR: δ = 2.4 (s, *Me*), 3.8 (s, *OMe*), 5.0 (s, CH₂), 5.2 (s, CH₂), 5.6–5.8 (br s, NH₂), 6.8–7.4 (m, 13 *ar* H, OH), 11.9–12.1 (br s, NH) ppm.

2-Amino-N-benzyl-2-hydroxyimino-N-(methoxycarbonyl)-1-(4-methylphenylhydrazono)ethane-1-sulfonamide (27c, $C_{18}H_{21}N_5O_5S$)

From **20e** (3.8 g, 10 mmol) and 1.4 g H₂NOH×HCl as described for **11a**. Yield 3.8 g (90%); mp 155°C (*Me*OH); IR: $\bar{\nu} = 3500, 3250$ (NH), 3030, 2960, 2920, 2860 (CH), 1730 (CO), 1350, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 2.4$ (s, *Me*), 3.71 (s, OM*e*), 5.06 (s, CH₂), 5.8–6.1 (br s, NH₂), 7.0–7.6 (m, 9 *ar* H, OH). 13.9–14.1 (br s, NH) ppm.

$\label{eq:2-Amino-N-benzyl-N-(benzyloxycarbonyl)-2-hydroxyimino-1-phenylhydrazonoethane-1-sulfonamide~(27d, C_{23}H_{23}N_5O_5S)$

From **20c** (4.5 g, 10 mmol) and 1.4 g H₂NOH×HCl as described for **11a**. Purification by CC (CHCl₃/ *AcOEt* 20/1). Yield 0.48 g (10%); mp 133–136°C (CHCl₃/*AcOEt*); IR: $\bar{\nu}$ = 3480–3260 (NH₂, NH, OH), 3060, 3020, 2970 (CH), 1730 (CO), 1330, 1150 (SO₂) cm⁻¹; ¹H NMR: δ = 5.15 (s, CH₂), 5.22 (s, CH₂), 5.6–5.8 (br s, NH₂), 6.0–7.0 (br s, OH), 7.0–7.5 (m, 15 *ar* H), 12.5–12.7 (br s, NH) ppm.

$\label{eq:constraint} \begin{array}{l} 5\text{-}Amino\text{-}N\text{-}benzyl\text{-}N\text{-}(benzyloxycarbonyl)\text{-}2\text{-}(4\text{-}methylphenyl)\text{-}2H\text{-}1,2,3\text{-}triazol\text{-}4\text{-}sulfonamide} \\ \textbf{(28a, C_{24}H_{23}N_5O_4S)} \end{array}$

At 0°C, **27a** (1.0 g, 2 mmol) and 30 cm³ POCl₃ were stirred for 1 h, the mixture was concentrated *in vacuo*, the residue was dissolved in CHCl₃, washed with an aqueous NaHCO₃ solution, dried (Na₂SO₄), and the solvent was evaporated *in vacuo*. Yield 0.95 g (90%); mp 168–169°C (*Me*OH); IR: $\bar{\nu} = 3380$, 3280 (NH₂), 3030, 2990 (CH), 1730 (CO), 1630 (C=N), 1380, 1170 (SO₂) cm⁻¹; ¹H NMR: $\delta = 2.35$ (s, *Me*), 5.02 (s, CH₂), 5.16 (s, CH₂), 5.3–5.5 (br s, NH₂), 7.0–7.5 (m, 14 *ar* H) ppm.

5-Amino-N-(benzyloxycarbonyl)-N-(4-methoxybenzyl)-2-(4-methylphenyl)-2H 1 2 2 triazol 4 culturaride (28b C H N O S)

$\it 2H-1,2,3-triazol-4-sulfonamide~(\textbf{28b},~C_{25}H_{25}N_5O_5S)$

From **27b** (2.2 g, 4 mmol) as described for **28a**. Yield 1.7 g (80%); mp 157–158°C (*Me*OH); IR: $\bar{\nu}$ = 3460 (NH₂), 3040, 2980 (CH), 1730 (CO), 1635 (C=N), 1385, 1165 (SO₂) cm⁻¹; ¹H NMR: δ = 2.42 (s, *Me*), 3.75 (s, OM*e*), 5.00 (s, CH₂), 5.15 (s, CH₂), 5.2–5.4 (br s, NH₂), 6.7–7.5 (m, 13 *ar* H) ppm.

$\label{eq:2.1} 5-Amino-N-benzyl-N-(methoxycarbonyl)-2-(4-methylphenyl)-2H-1,2,3-triazol-4-sulfonamide \mbox{(28c, $C_{18}H_{19}N_5O_4S$)}$

From **27c** (4.2 g, 10 mmol) as described for **28a**. Yield 3.6 g (90%); mp 142°C (*Me*OH); IR: $\bar{\nu} = 3400$ (NH₂), 3020, 2920 (CH), 1730 (CO), 1630 (C=N), 1380, 1170 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 2.36$ (s, *Me*), 3.66 (s, *OMe*), 5.03 (s, CH₂), 6.2–6.4 (br s, NH₂), 7.2–7.6 (m, 9 *ar* H) ppm.

$\label{eq:2.1} \begin{array}{l} 5\text{-}Amino\text{-}N\text{-}benzyl\text{-}N\text{-}(benzyloxycarbonyl)\text{-}2\text{-}phenyl\text{-}2H\text{-}1,2,3\text{-}triazol\text{-}4\text{-}sulfonamide} \\ \textbf{(28d, $C_{23}H_{21}N_5O_4S$)} \end{array}$

From **27d** (1.0 g, 2 mmol) as described for **28a**. Yield 0.9 g (90%); mp 120°C (*Me*OH); IR: $\bar{\nu} = 3450$ (NH₂), 3060, 3030, 2950 (CH), 1730 (CO), 1630 (C=N), 1380, 1160 (SO₂) cm⁻¹; ¹H NMR: $\delta = 5.01$ (s, CH₂), 5.12 (s, CH₂), 5.7–5.9 (br s, NH₂), 7.1–7.5 (m, 15 *ar* H) ppm.

5-Amino-N-benzyl-2-(4-methylphenyl)-2H-1,2,3-triazol-4-sulfonamide (**29a**, $C_{16}H_{17}N_5O_2S$) A solution of 0.2 g Na in 10 cm³ *Et*OH was added to a solution of **28a** (0.5 g, 1 mmol) in 20 cm³ *Et*OH. The mixture was stirred at room temperature for 1 h, neutralized with HCl (36%), evaporated *in vacuo*,

and the residue was purified by CC (CHCl₃/*AcOEt*). Yield 0.17 g (50%); mp 103–105°C (*MeOH*); IR: $\bar{\nu} = 3450, 3280$ (NH), 3020, 2830 (CH), 1630 (C=N), 1340, 1160 (SO₂) cm⁻¹; ¹H NMR: $\delta = 2.35$ (s, *Me*), 4.3 (d, *J* = 7 Hz, CH₂), 5.4–5.6 (br s, NH₂), 6.0–6.1 (t, *J* = 7 Hz, NH), 7.2–7.4 (m, 9 ar H) ppm; MS (70 eV): m/z (%) = 344 (1.57, M⁺), 106 (100, C₇H₈N).

$\label{eq:2.1} \begin{array}{l} 5\text{-}Amino\text{-}N\text{-}(4\text{-}methoxybenzyl)\text{-}2\text{-}(4\text{-}methylphenyl)\text{-}2H\text{-}1,2,3\text{-}triazol\text{-}4\text{-}sulfonamide} \\ \textbf{(29b, }C_{17}H_{19}N_5O_3S\textbf{)} \end{array}$

From **28a** (0.5 g, 1 mmol) as described for **29a**. Yield 0.21 g (60%); mp 183–184°C (CHCl₃/ cyclohexane); IR: $\bar{\nu} = 3490$, 3260 (NH), 3030, 2830 (CH), 1635 (C=N), 1350, 1165 (SO₂) cm⁻¹; ¹H NMR: $\delta = 2.42$ (s, *Me*), 3.72 (s, *OMe*), 4.4 (d, J = 7 Hz, CH₂), 5.9–6.0 (br s, NH₂), 6.6–7.3 (m, 8 *ar* H, NH) ppm; MS (70 eV): m/z (%) = 372 (3.8, M⁺ – 1); 136 (100, C₈H₁₀NO).

5-Amino-N-benzyl-2-phenyl-2H-1,2,3-triazol-4-sulfonamide (29c, C₁₅H₁₅N₅O₂S)

From **28d** (0.46 g, 1 mmol) as described for **29a**. Yield 0.16 g (50%); mp 120–121°C (CHCl₃/cyclocyclohexane); IR: $\bar{\nu} = 3450-3350$ (NH), 3020 (CH), 1620 (C=N), 1340, 1160 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 4.4$ (d, J = 6 Hz, CH₂), 6.0–6.3 (br s, NH₂), 7.0–7.6 (m, 10 *ar* H, NH) ppm.

$\label{eq:last_loss} 2-Acetoximino-2-amino-N-benzyl-N-(benzyloxycarbonyl)-1-phenylhydrazonoethane-benzyl-N-(benzyloxycarbonyl-N-(benzyloxycarbonyl-N-(benzyloxycarbonyl-N-(benzyloxycarbonyl-N-(benzyloxycarbonyl-N-(benzyloxycarbonyl-N-(benzyloxycarbonyl-N-(benzyloxycarbonyl-N-(benzyloxycarbonyl-N$

1-sulfonamide (**30**, C₂₅H₂₅N₅O₆S)

Compound **27d** (0.5 g, 1 mmol) in 20 cm³ Ac_2O was refluxed for 1 h. The mixture was evaporated *in vacuo*, the residue was dissolved in CHCl₃, washed with a NaHCO₃ solution, dried (Na₂SO₄), and evaporated *in vacuo*. Yield 0.25 g (50%); mp 84°C (*Me*OH); IR: $\bar{\nu} = 3480$, 3380, 3260 (NH₂, NH), 3060, 3030 (CH), 1760, 1730 (CO), 1620 (C=N), 1530 (C=N-NH), 1340, 1160 (SO₂) cm⁻¹; ¹H NMR: $\delta = 2.17$ (s, *Me*), 5.04 (s, CH₂), 5.11 (s, CH₂), 5.8–6.0 (br s, NH₂), 7.0–7.5 (m, 15 *ar* H), 13.4–13.6 (s, NH) ppm.

$\label{eq:second} 5-Acetylamino-N-benzyl-N-(benzyloxycarbonyl)-2-(4-methylphenyl)-2H-benzyl-N-(benzyloxycarbonyl)-2-(4-methylphenyl)-2H-benzyloxycarbonyl)-2-(4-methylphenyl)-2H-benzyloxycarbonyl)-2-(4-methylphenyl)-2H-benzyloxycarbonyl)-2-(4-methylphenyl)-2H-benzyloxycarbonyl)-2-(4-methylphenyl)-2H-benzyloxycarbonyl)-2-(4-methylphenyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyl)-2H-benzyloxycarbonyloxy$

1,2,3-triazol-4-sulfonamide (**31**, C₂₆H₂₅N₅O₅S)

Compound **28a** (0.47 g, 1 mmol), 20 cm³ AcOH, and 20 cm³ Ac₂O were refluxed for 2 h, cooled to room temperature, and evaporated *in vacuo*. The residue was dissolved in CHCl₃, washed with an aqueous NaHCO₃ solution, dried (Na₂SO₄), and evaporated *in vacuo*. Yield 0.4 g (80%); mp 157°C (*Me*OH); IR: $\bar{\nu}$ = 3380, 3280 (NH), 3060, 3030 (CH), 1730 (CO), 1370, 1165 (SO₂) cm⁻¹; ¹H NMR: δ = 2.13 (s, *Me*), 2.41 (s, *Me*), 5.0–5.2 (2s, 2 CH₂), 5.4–5.6 (br s, NH), 7.0–7.5 (m, 14 *ar* H) ppm.

$\label{eq:2.1} 5-(S-Ethylthiocarbonylamino)-N-(4-methoxybenzyl)-2-(4-methylphenyl)-2H-1,2,3-triazol-4-sulfonamide~({\bf 32},~C_{20}H_{23}N_5O_4S_2)$

Under N₂ at -78° C, 0.65 cm³ BuLi were added to **29b** (0.35 g, 1 mmol) in 50 cm³ THF, then, after 10 min S-ethyl chlorothioformate (0.12 g, 1 mmol) in 10 cm³ THF was added, and after stirring for 1 h 0.2 cm³ BuLi were added. The mixture was warmed to room temperature, washed with a satd NaCl solution, dried (MgSO₄), and evaporated *in vacuo*. Yield 0.35 g (80%); mp 165–167°C (*MeOH*); IR: $\bar{\nu} = 3420$ (NH₂), 1680 (COSEt), 2960, 2920 (CH), 1630 (C=N), 1370, 1155 (SO₂) cm⁻¹; ¹H NMR: $\delta = 1.0-1.3$ (t, J = 7 Hz, *Me*), 2.39 (s, *Me*), 2.7–3.0 (q, J = 7 Hz, CH₂), 3.75 (s, OMe), 5.05 (s, CH₂), 5.6 (br s, 2 NH), 6.8–7.5 (m, 8 *ar* H) ppm.

4-Amino-N-benzylpyrimidine-5-sulfonamide (**33a**, C₁₁H₁₂N₄O₂S)

Compound **2c** (2.1 g, 10 mmol) and formamidinium acetate (6.3 g, 60 mmol) in 50 cm³ *Bu*OH were refluxed for 6 h, the mixture was cooled to room temperature, the precipitate was separated, and the filtrate was concentrated yielding more precipitate. The combined precipitates were crystallized. Yield 2.1 g (80%); mp 210°C (*THF*); IR: $\bar{\nu} = 3420$, 3300 (NH), 3120, 3020, 2920 (CH), 1650 (C=N), 1320, 1130 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 4.2$ (s, CH₂), 7.2–7.5 (m, 5 *ar* H, NH₂), 8.3–8.5 (m, 2 *ar* H, NH) ppm.

4-Amino-N-(4-methoxybenzyl)pyrimidine-5-sulfonamide (33b, C₁₂H₁₄N₄O₃S)

From **2d** (2.4 g, 10 mmol) as described for **33a**. Yield 2.35 g (80%); mp 197–199°C (*THF*); IR: $\bar{\nu} = 3430, 3300$ (NH), 3120, 2990 (CH), 1660 (C=N), 1330, 1170 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 3.9$ (s, *OMe*), 4.2 (s, CH₂), 3.9–5.0 (br s, NH₂), 6.8–7.5 (m, 4 *ar* H), 7.0–8.0 (br s, NH), 8.50, 8.51 (2s, 2 *ar* H) ppm.

4-Amino-N-phenylpyrimidine-5-sulfonamide (33c, C₁₀H₁₀N₄O₂S)

From **2b** (2.0 g, 10 mmol) as described for **33a**. Yield 2.0 g (80%); mp >250°C (dec, *THF*); IR: $\bar{\nu} = 3430$, 3310 (NH₂, NH), 3160, 3060, 3020, 2950, 2880 (CH), 1650 (C=N), 1330, 1140 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 7.0-7.4$ (s, 5 *ar* H), 6.5–8.5 (br s, NH₂), 8.5 (s, 2 *ar* H), 10.3–10.5 (br s, NH) ppm.

4-Amino-N-(4-methoxyphenyl)pyrimidine-5-sulfonamide (33d, C₁₁H₁₂N₄O₂S)

From **2f** (2.1 g, 10 mmol) as described for **33a**. Yield 2.3 g (80%); mp 291°C (*THF*); IR: $\bar{\nu} = 3490$, 3290 (NH₂), 3120, 2995, 2900 (CH), 1640 (C=N), 1320, 1160 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆): $\delta = 3.65$ (s, *OMe*), 6.8–7.2 (m, 4 *ar* H), 8.30, 8.45 (2s, 2 *ar* H), 9.0–6.5 (br s, NH₂), 10.0 (br s, NH) ppm.

4-Amino-N-propylpyrimidine-5-sulfonamide (33e, C₇H₁₂N₄O₂S)

From **2g** (1.6 g, 10 mmol) as described for **33a**. Purification by CC (*AcOEt*) yielded 0.1 g (5%) **35** as a by-product. Yield 1.5 g (70%); mp 164°C (*AcOEt*); IR: $\bar{\nu} = 3430, 3300$ (NH₂), 3120, 3020, 2960 (CH), 1650 (C=N), 1325, 1160 (SO₂) cm⁻¹; ¹H NMR (acetone-d₆): $\delta = 1.0$ (m, *Me*), 1.4–1.8 (m, CH₂), 2.9–3.2 (m, CH₂), 6.3–7.5 (br s, NH₂, NH), 8.4–8.5 (2s, 2 *ar* H) ppm.

4-Amino-N-(2-chlorobenzyl)pyrimidine-5-sulfonamide (33f, C₁₁H₁₁ClN₄O₂S)

From **2h** (2.4 g, 10 mmol) as described for **33a**. Yield 2.2 g (80%); mp 222°C (*Me*OH); IR: $\bar{\nu} = 3411$, 3319 (NH), 1657 (C=N), 1331, 1135 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆, 400 MHz): $\delta = 4.15$ (s, CH₂), 7.2–7.5 (m, 4 *ar* H), 8.40, 8.45 (2s, 2 *ar* H) ppm; MS (70 eV): m/z (%) = 299 (80, M⁺).

4-Amino-N-(4-chlorobenzyl)pyrimidine-5-sulfonamide (**33g**, C₁₁H₁₁ClN₄O₂S)

From **2i** (2.4 g, 10 mmol) as described for **33a**. Yield 2.3 g (80%); mp 244°C (*Me*OH); IR: $\bar{\nu} = 3411$, 3319 (NH), 1657 (C=N), 1331, 1135 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆, 90 MHz): $\delta = 4.1$ (s, CH₂), 7.1–7.5 (m, 4 *ar* H), 8.4, 8.5 (2s, 2 *ar* H) ppm.

$\label{eq:2.1} 4-Amino-N-(4-fluorobenzyl) pyrimidine-5-sulfonamide~~({\bf 33h},~C_{11}H_{11}FN_4O_2S)$

From **2k** (2.3 g, 10 mmol) as described for **33a**. Yield 2.2 g (80%); mp 197°C (*Me*OH); IR: $\bar{\nu} = 3400$, 3300 (NH), 1650 (C=N), 1330, 1135 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆, 90 MHz): $\delta = 4.1$ (s, CH₂), 6.97–7.4 (m, 4 ar H), 8.4, 8.5 (2s, 2 ar H) ppm.

4-Amino-N-(4-bromophenyl)pyrimidine-5-sulfonamide (33i, C10H9BrN4O2S)

From **2l** (2.7 g, 10 mmol) as described for **33a**. Yield 2.0 g (60%); mp 300°C (*Me*OH); IR: $\bar{\nu} = 3436$, 3305 (NH), 1650 (C=N), 1329, 1135 (SO₂) cm⁻¹; ¹H NMR (*DMSO*-d₆, 90 MHz): $\delta = 7.0-7.6$ (m, 4 ar H), 8.5 (s, 2 ar H) ppm.

4-Acetylamino-N-benzylpyrimidine-5-sulfonamide (34a, C₁₃H₁₄N₄O₃S)

Compound **33a** (1.3 g, 5 mmol), 20 cm³ AcOH, and 20 cm³ Ac₂O were refluxed for 3 h, cooled to room temperature, and evaporated *in vacuo*. The residue was dissolved in CHCl₃, washed with an aqueous AcONa solution, dried (Na₂SO₄), and the solvent was evaporated *in vacuo*. Yield 1.0 g (65%); mp 218°C (CHCl₃/CCl₄); IR: $\bar{\nu}$ = 3360 (NH), 3120, 2950 (CH), 1650 (C=N), 1320, 1140 (SO₂) cm⁻¹; ¹H NMR: δ = 2.4 (s, *Me*), 3.2–3.6 (br s, NH), 4.2 (s, CH₂), 7.2 (s, 5 *ar* H), 8.8–9.3 (m, 2 *ar* H, NH) ppm.

4-Acetylamino-N-acetyl-N-(4-chlorobenzyl)pyrimidine-5-sulfonamide (**34b**, $C_{15}H_{15}ClN_4O_4S$) From **33g** (1.5 g, 5 mmol) as described for **34a**. Yield 0.8 g (42%); mp 180°C (acetone); IR: $\bar{\nu} = 3341$ (NH), 1703 (CO), 1571 (C=N), 1374, 1147 (SO₂) cm⁻¹; ¹H NMR (90 MHz): $\delta = 2.2$ (s, *Me*), 2.5 (s, *Me*), 5.1 (s, CH₂), 7.1–7.5 (m, 4 ar H), 8.7, 9.0 (2s, 2 ar H), 9.6 (s, NH) ppm.

4-Acetylamino-N-acetyl-N-phenylpyrimidine-5-sulfonamide (**34c**, C₁₄H₁₄N₄O₄S)

From **33c** (1.3 g, 5 mmol) as described for **34a**. Yield 0.5 g (30%); mp 160°C (acetone); IR: $\bar{\nu} = 3345$ (NH), 1710 (CO), 1570 (C=N), 1375, 1150 (SO₂) cm⁻¹; ¹H NMR: $\delta = 1.9$ (s, *Me*), 2.5 (s, *Me*), 7.2–7.7 (m, 5 *ar* H), 8.9, 9.1 (2s, 2 *ar* H), 9.7 (s, NH) ppm.

2-Amino-N-propyl-1-cyanoethene-1-sulfonamide (**35**, C₆H₁₁N₃O₂S) Compound **35** was isolated as by-product from the synthesis of **33e**. Yield 0.1 g (5%); IR: $\bar{\nu}$ = 3440, 3280 (NH), 2200 (CN), 1650 (C=C), 1310, 1160 (SO₂) cm⁻¹.

(*RS*)-2-(4-Chlorobenzyl)-3-ethoxy-3,4-dihydro-2H-pyrimido[4,5-e][1,2,4]thiadiazine 1,1-dioxide (**36a**, C₁₄H₁₅ClN₄O₃S)

Compound **33g** (0.3 g, 1 mmol), 2 cm³ HC(O*Et*)₃, and 3 drops *Ac*OH were refluxed for 6 h, cooled to room temperature, and the precipitate was isolated. Yield 0.2 g (56%); mp 179°C (*Me*OH); IR: $\bar{\nu} = 3180$, 2983 (CH), 1593 (C=N), 1346, 1176 (SO₂), 1121 (C-O) cm⁻¹; ¹H NMR (*DMSO*-d₆, 90 MHz): $\delta = 1.1$ (t, J = 5.8 Hz, *Me*), 3.6 (q, J = 11.6 Hz, CH₂), 4.3 (d, J = 5.8 Hz, CH₂), 5.96 (s, 3-H), 7.4 (s, 4 *ar* H), 8.7, 8.8 (2s, 2 *ar* H), 9.6 (br s, NH) ppm.

(*RS*)-2-(4-Fluorobenzyl)-3-ethoxy-3,4-dihydro-2H-pyrimido[4,5-e][1,2,4]thiadiazine 1,1-dioxide (**36b**, C₁₄H₁₅FN₄O₃S)

From **33h** (0.3 g, 1 mmol) as described for **36a**. Yield 0.22 g (65%); mp 170°C (*Me*OH); IR: $\bar{\nu} = 3179$, 2984 (CH), 1593 (C=N), 1345, 1182 (SO₂), 1125 (C–O) cm⁻¹; ¹H NMR (*DMSO*-d₆, 300 MHz): $\delta = 1.03$ (t, J = 6.6 Hz, *Me*), 3.56 (q, J = 9.9 Hz, CH₂), 4.2 (d, CH₂), 5.8 (s, 3-H), 7.1–7.3 (m, 4 *ar* H), 8.70, 8.74 (2s, 2 *ar* H), 9.6 (br s, NH) ppm.

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